



Co-Optimization of Fuels & Engines

FOR TOMORROW'S ENERGY-EFFICIENT VEHICLES

Co-Optima Fuel Property Team

Part 1 - Fuel Properties and Chemical Kinetics

June 9, 2016



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Fuel Property Team

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1. National Renewable Energy Laboratory
2. Oak Ridge National Laboratory
3. Pacific Northwest National Laboratory
4. Sandia National Laboratories
5. Lawrence Livermore National Laboratory
6. Argonne National Laboratory

Co-Optima DOE VTO Management Team: Kevin Stork and Gurpreet Singh



Overview: Fuel Properties and Chemical Kinetics

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Central Fuel Properties Hypothesis

Development of Fuel Property Database (Fioroni in collaboration with LGHG Fuels Team)

Development of Fuel Screening Criteria – Thrust I (McCormick, Szybist, Miles in collaboration with LGHG Fuels Team)

Heat of Vaporization Measurement (Fioroni)

Measurement of Autoignition Properties with Small Volumes
(Fioroni/McCormick, Goldsborough, McNenly)

Fuel Structure-Property Correlations (Bays)

Fuel Property Blending Models (McCormick)

Kinetic Mechanism Development (Pitz, Zigler)



Milestones: Fuel Properties and Chemical Kinetics

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Tracked Milestones:

Analysis of up to 20 molecules that represent as many functional groups as possible which will increase our capacity to understand how chemical properties (functional groups) correlate to needed fuel properties and materials compatibility. (Owner NREL, due FY16 Q2) – **Complete**

Provide up to five fuels and 4 blending levels for blend determination studies. (Owner NREL, due FY16 Q4) – **On Track**

Provide a report to DOE and Optima documenting the current development status and potential for each of the three small volume autoignition testing techniques. (Owner LLNL, due FY16 Q4) – **On Track**

Develop a preliminary chemical kinetic mechanism for anisole, a compound relevant to pyrolysis oil-derived fuels. (Owner LLNL, due FY16 Q4) – **On Track**



Central Fuel Properties Hypothesis

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If we correctly identify the critical fuel properties that affect efficiency and emissions performance for downsized, boosted SI engines, then fuels that have those properties will provide optimal engine performance.

Also applies to Thrust II advanced compression ignition engines, but at a much lower level of development.

- With the correct properties, performance is a function of properties not composition or molecular structure
- Hypothesis may be true or very nearly true for petroleum-derived fuels
- Tested using engine experiments on biomass-derived fuels (oxygenates, targeted hydrocarbons) with fuel property values beyond the range exhibited by today's petroleum-derived fuels.



NREL (Fioroni) \$100k: Development of Fuel Property Database (BETO)

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Objectives

- Create a publically-accessible Fuel Property Database for conventional new candidate fuels

Approach

- Filemaker Pro platform with on-line access
- Fuel molecules and mixtures proposed by LGHG Fuels Team
- Data measured or from literature

Accomplishment

- 366 pure compound and 12 mixtures – so far
- Fields for 25 specific fuel properties
- Used extensively in preparing Q2 Co-Optima Milestone Report (BETO)

Found Pure Compound Correct or Update this record

IUPAC name:

Synonyms:

Molecular Weight:

Molecular Formula:

CAS#:

Functional Group:

Drop an image of the Structure here:

OCC(O)CCC

Popover

SEARCH PROPERTIES

Both "pure" IUPAC compoundname "methyl" "Point" all records with and for "blend" will be searched. (L)

Boiling "0 ... 14" finds both "methyl" in the name AND a boiling point range between 0 and 14 will be

Molecular Weight:

Molecular Formula:

CAS#:

Safety

LFL, LEL, (%): UFL, UEL, (%):

Flash Point (°C): Autoignition Temp (°C):

Peroxide Former:

Health

Rat Oral LD50 (mg/kg):

Properties

Melting Point (°C): <input type="text"/>	Boiling Point (°C): <input type="text"/>	Peroxide Value: <input type="text"/>	T90 (°C): <input type="text"/>
Cloud Point (°C): <input type="text"/>	BP (°C): <input type="text"/>	T95 (°C): <input type="text"/>	
Density (g/cm ³): <input type="text"/>	Heat of Vaporization (kJ/mol): <input type="text"/>	FBP (°C): <input type="text"/>	Surface Tension (dyne/cm): <input type="text"/>
Viscosity (cSt): <input type="text"/>	Vapor Pressure (kPa): <input type="text"/>	Corrosion: <input type="text"/>	PMI: <input type="text"/>
MON: <input type="text"/>	RON: <input type="text"/>	Lubricity: <input type="text"/>	
UNF: <input type="text"/>	DCN: <input type="text"/>	Stability: <input type="text"/>	Functional Group: <input type="text"/>
Critical Pressure (kPa): <input type="text"/>	Critical Temperature (K): <input type="text"/>	Oxidation Stability: <input type="text"/>	Thermal Stability: <input type="text"/>
Acentric Factor: <input type="text"/>	Acid Value: <input type="text"/>	Water Solubility (mg/L): <input type="text"/>	Dispersion: <input type="text"/>



NREL (McCormick) \$75k: Development of Fuel Screening Criteria (BETO)

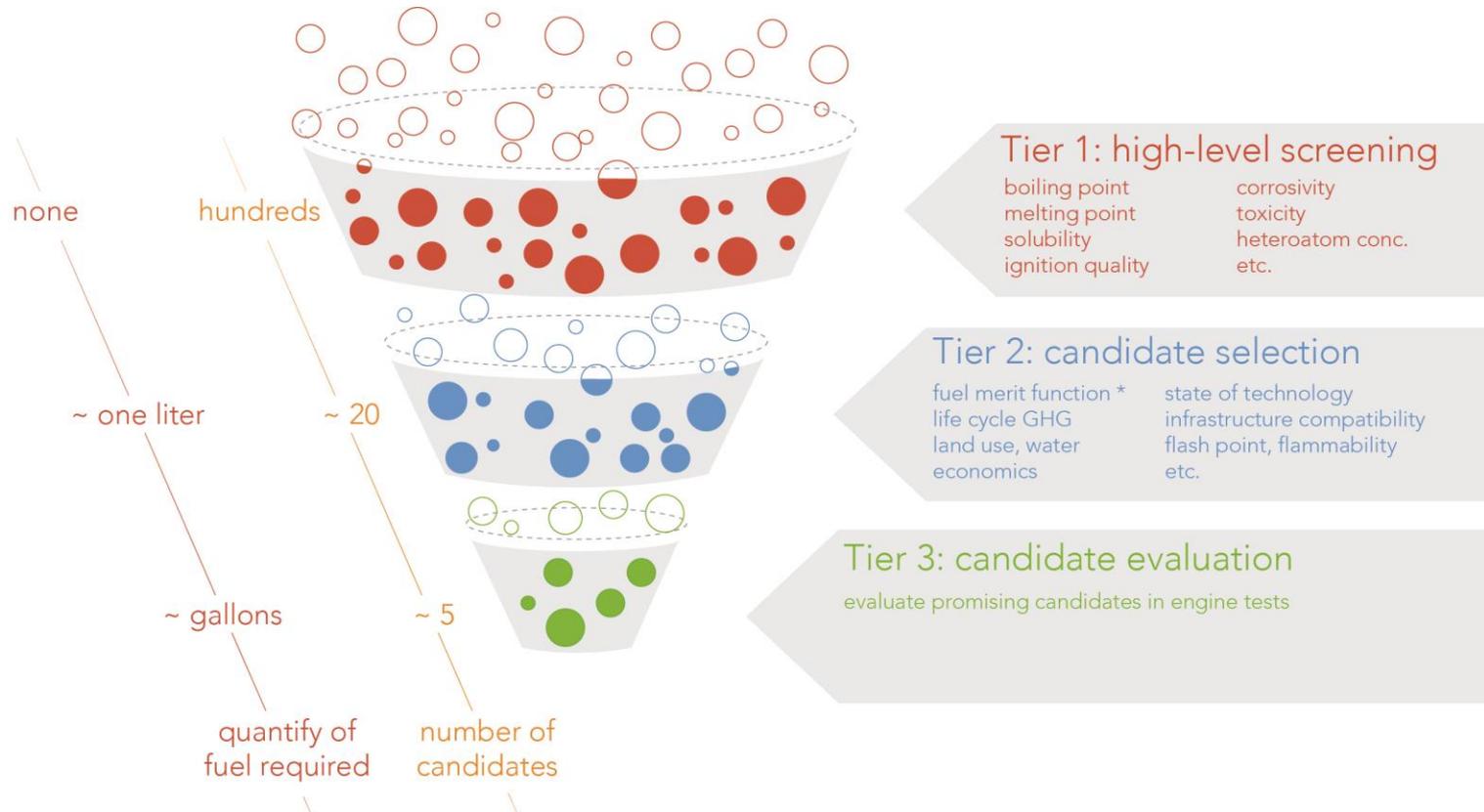
7

Objectives

- Develop strategy for screening large numbers of candidate bio-blendstocks to select most promising based on fuel properties

Approach

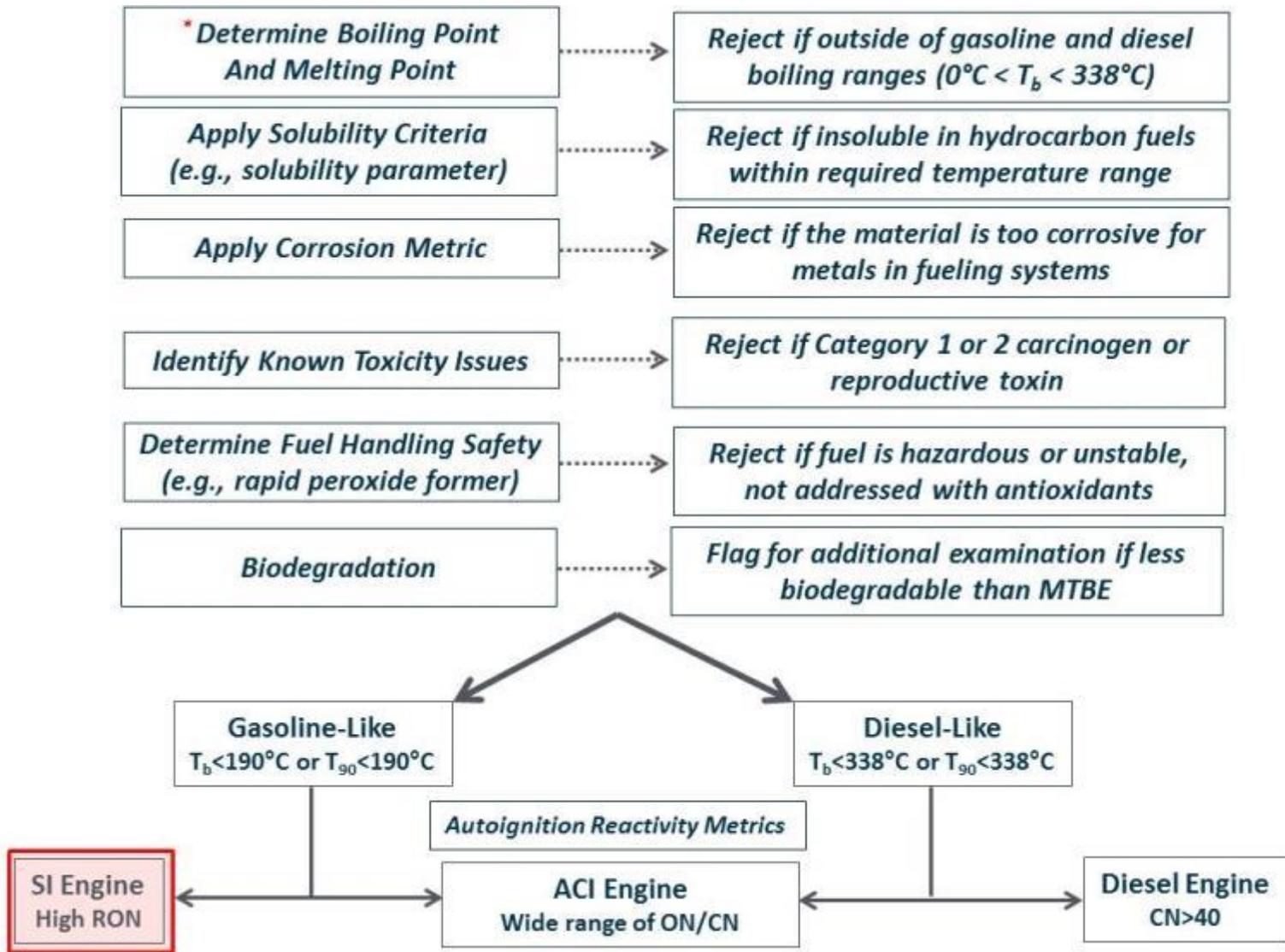
- Three tier evaluation protocol





Tier 1 Screening for Thrust I Candidates

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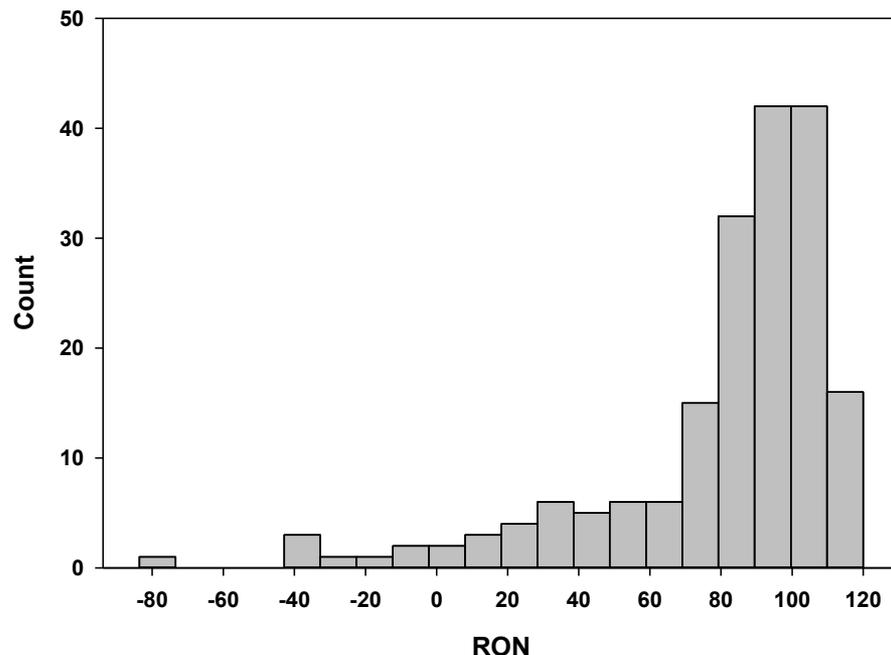


Accomplishment: Selection of Most Promising Thrust I Bio-Blendstocks

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For Thrust I (advanced SI engines) apply Tier 1 Criteria:

- Melting point/cloud point below -10°C
- Boiling point between 20°C and 165°C
- Measured or estimated RON ≥ 98
- Meet toxicity, corrosion, solubility, and biodegradation requirements
- 34 promising bio-blendstocks from all functional group classes – *strong matrix for testing central fuel properties hypothesis*



Database Future Developments

- Fully populate property data for most promising candidates
- Add capability for data on finished fuel blends

2-pentanol	6032-29-7	99.4 ^a	Biochemical
Alkanes			
High-Octane Gasoline Blendstock- triptane rich		108 ^d	Thermochemical
Alkenes			
4-methylene-1-(1-methylethyl)bicyclo[3.1.0]hexane (sabinene)	3387-41-5	107 ^c	Biochemical
Aromatics			
1,3,5-trimethylbenzene (mesitylene)	108-67-8	>120 ^a	Biochemical/Chemical
Vertifuel (60%+ aromatics)		105.7 ^d	Biochemical/Chemical
Esters			
Acetic acid, methyl ester	79-20-9	>120 ^a	Biochemical/Chemical
Butanoic acid, methyl ester	623-42-7	107.2 ^a	Biochemical/Chemical
Pentanoic acid, methyl ester	624-	99 ^b	Biochemical/Chemical
2-Methyl propanoic acid, methyl ester	547-63-7	103 ^c	Biochemical/Chemical
2-Methylbutanoic acid, methyl ester	868-57-5	110.5 ^a	Biochemical/Chemical
Acetic acid ethyl ester	141-78-6	118 ^a	Biochemical/Chemical

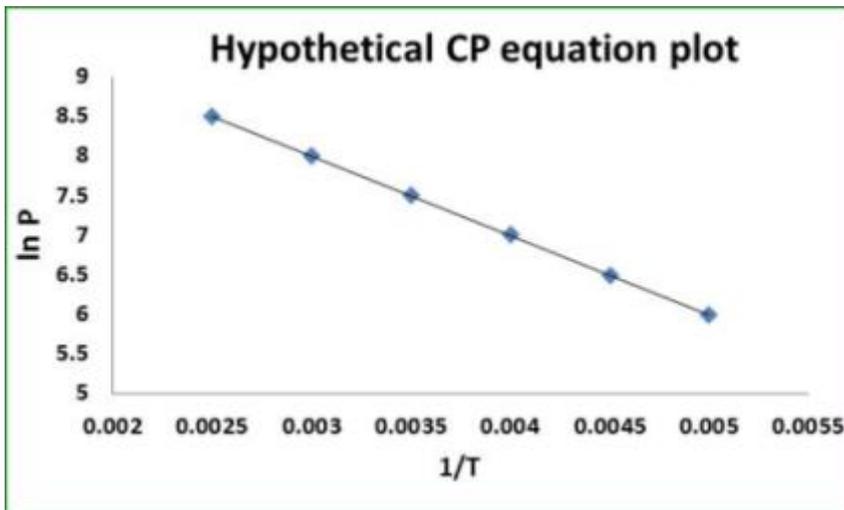


NREL (Fioroni) \$250k: Heat of Vaporization Measurement

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Objective

- Develop methods for measuring heat of vaporization (HOV) of complex mixtures
- Measure HOV as a function of fraction evaporated
- High HOV fuels may extend knock limit, reduce need for spark retard beyond knock limit, reduce need for rich operation



$$\ln[P_{T1,vap}/P_{T2,vap}] = (\Delta H_{vap}/R)[1/T_2 - 1/T_1]$$

P = Vapor pressure

T = temperature in Kelvin

ΔH_{vap} = HOV

R = ideal gas constant

- Pure compound HOV easily measured
 - Measure vapor pressure vs temperature and apply Clausius-Clapeyron equation
- Not applicable to blends such as gasoline

- *The vapor pressure is determined by the most volatile components of the fuel which make up only a small portion of the fuel*
- *This yields an unrealistically low value for HOV*



Fioroni: HOV Measurements

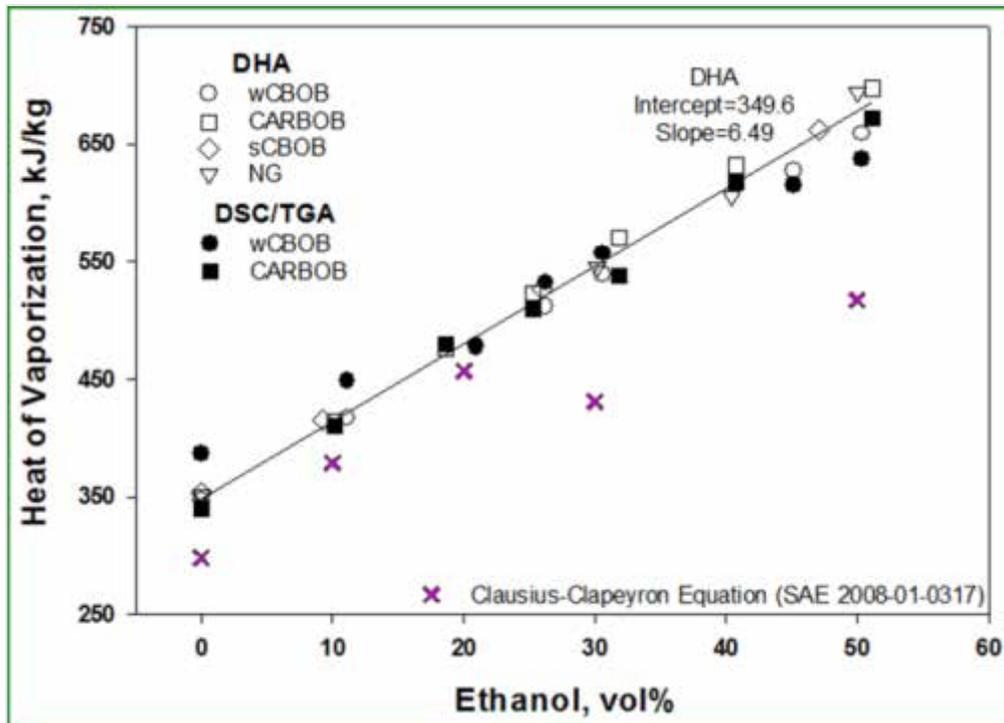
11

Approach

- Directly measure HOV by DSC/TGA
- Calculate HOV from detailed hydrocarbon analysis of gasoline (mole fraction weighted sum of pure component HOV)

Challenge: highly volatile samples evaporate as they are loaded into the instrument

Solution: Syringe injection of sample, larger sample pans, covered with pin-hole lids

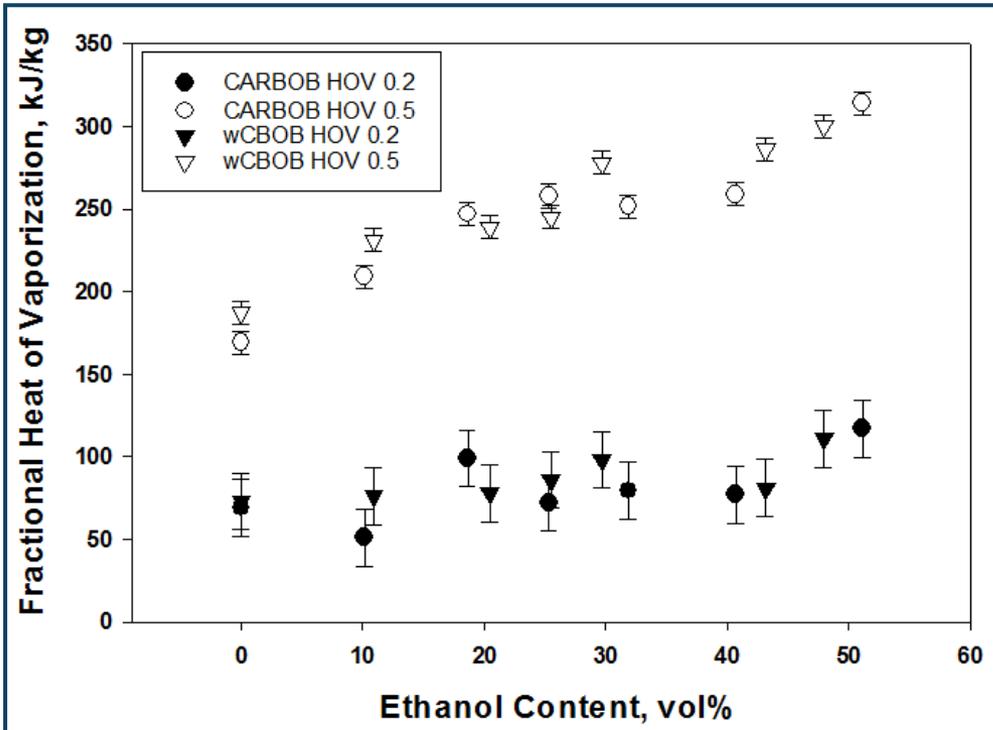


- DSC/TGA measurements agree well with DHA
- Values calculated using the CP equation fall below DSC/TGA or DHA, especially at high ethanol content
- Very similar HOV for wide range of gasolines and ethanol blends



Fioroni: HOV as a Function of Fraction Evaporated

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- Effect of ethanol on HOV20 may be negligible
- Increasing ethanol has significant effect on HOV50
- Working to improve precision of this measurement under project with CRC (AVFL-27)

Collaborations:

Coordinating Research Council
University of Colorado

HOV Future Work FY16/17

- Improve precision of DSC/TGA measurements
 - New dedicated instrument
 - Couple with much greater precision DSC measurements
- Couple with Advanced Distillation Curve experiments and DSC/TGA-MS to reveal composition as fuel evaporates
- Collaboration with NREL engine group on actual HOV effects

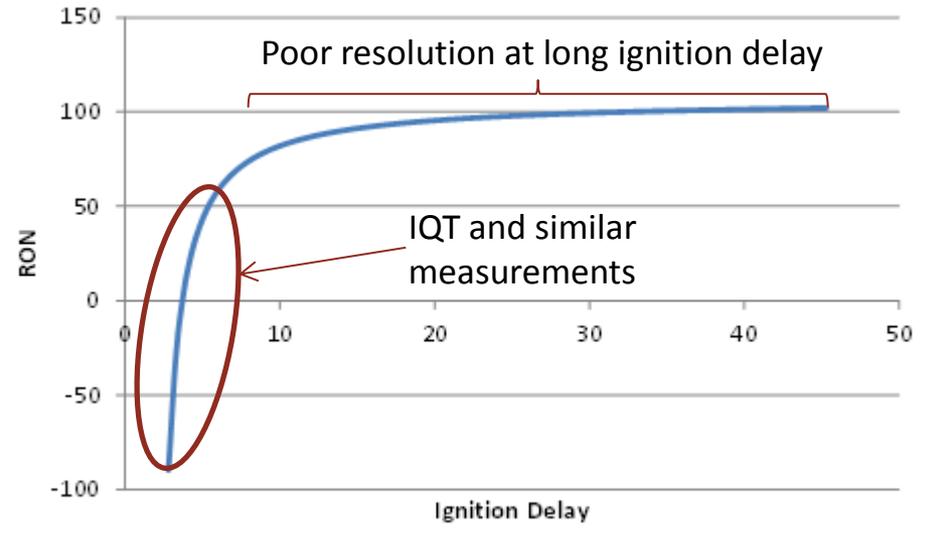


Measurement of Autoignition Properties with Small Volumes

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Background

- Providing early-stage feedback to identify promising fuel candidates, accelerate progress and eliminate effort and expense associated with unnecessary scale-up.
- Focused on using milliliter volume samples to characterize the autoignition behavior of fuels, the property that is the most problematic to predict.



Objectives

- Measure autoignition kinetics, or a variable correlated with common autoignition metrics such as RON or CN, using a very small volume of material

Three projects in various stages:

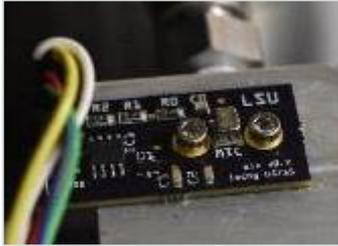
- LLNL (McNenly): Flames with Repetitive Extinction & Ignition
 - Ongoing Collaboration with Louisiana State University
- NREL (McCormick): Flow Reactor for Small Volume Autoignition Experiments
 - FY16 Start
- ANL (Goldsborough): RCM as a Small Volume Autoignition Experiment
 - FY16 Start



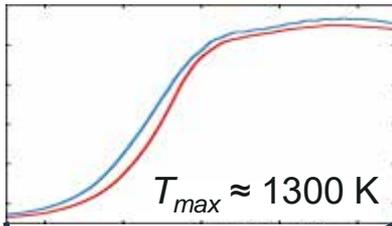
LLNL (McNenly) \$75k: μ FIT: Micro-liter Fuel Ignition Tester Flames with Repetitive Extinction & Ignition (FREI)

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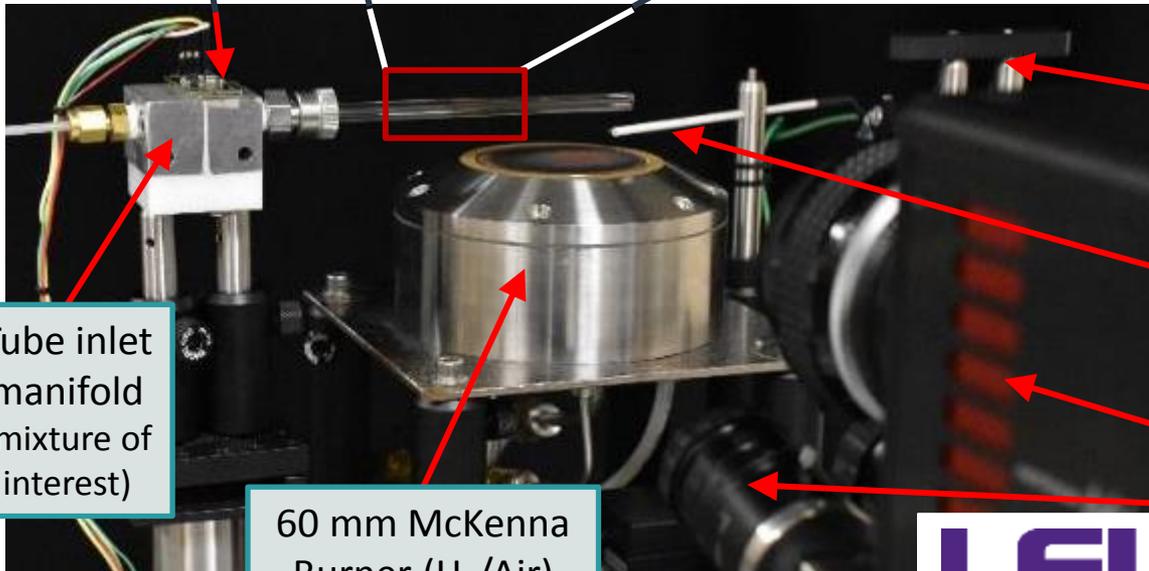
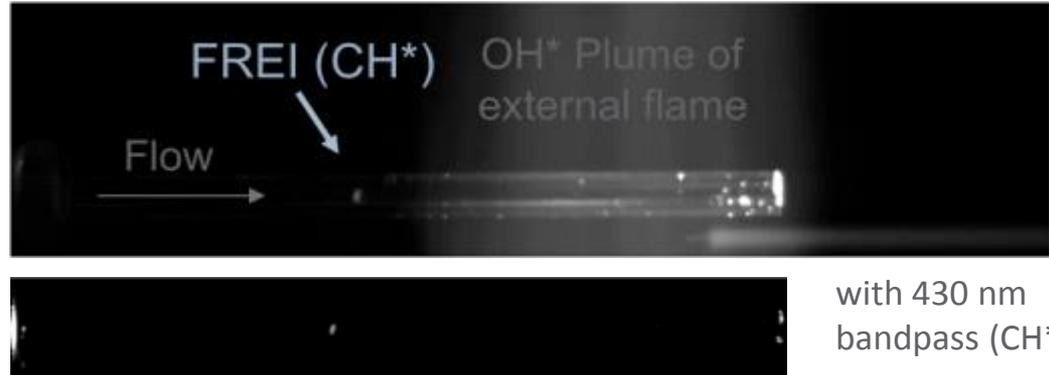
MEMS
microphone



Quartz Tube
(with imposed wall
temperature profile)



High Speed Videos (2000 fps; 2mm ID tube)



Tube inlet
manifold
(mixture of
interest)

60 mm McKenna
Burner (H_2/Air)
(external heating)

Translation stage for
wall temperature
characterization
(thermocouple not shown)

Monitoring
thermocouple

High-Speed and
Machine Vision Cameras



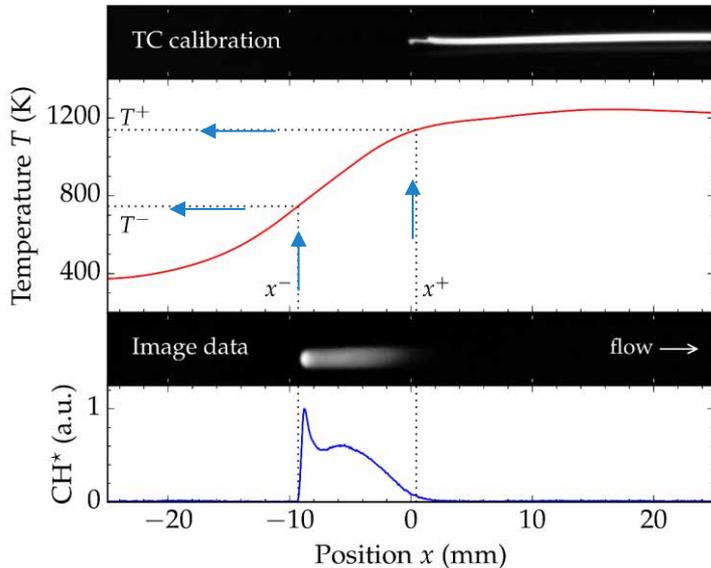


McNenly: μ FIT: Micro-liter Fuel Ignition Tester

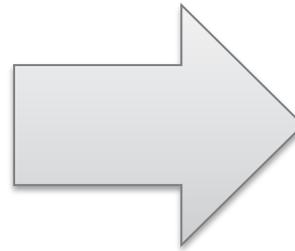
Flames with Repetitive Extinction & Ignition (FREI)

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TC data plus low-speed images yield temperatures



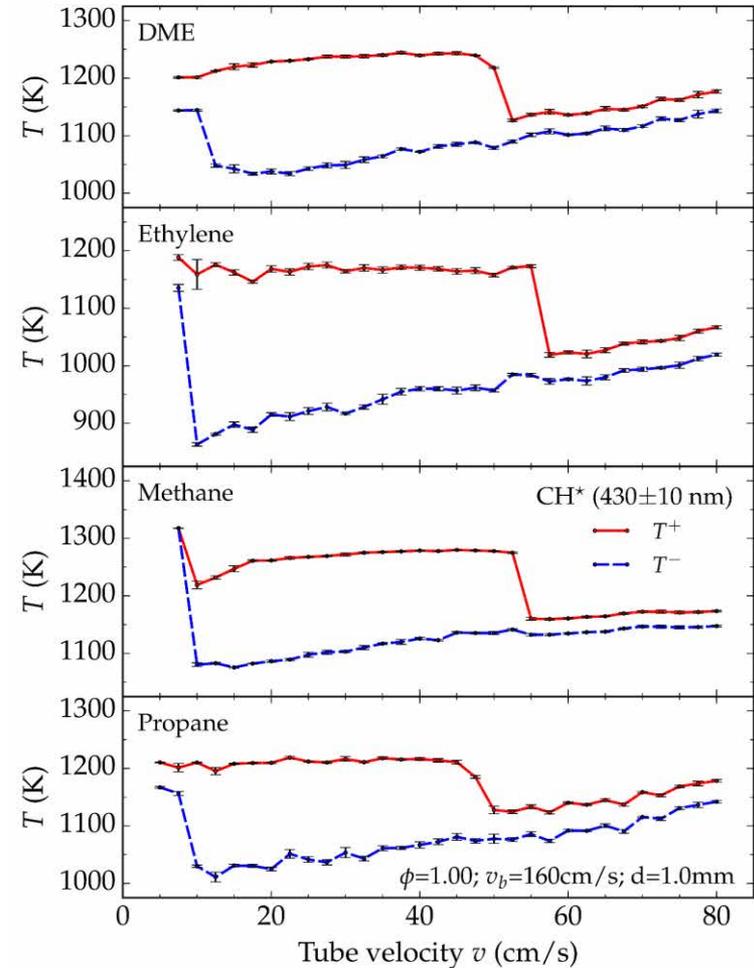
FREI characteristics (T^+ and T^-) depend on fuel



Fuel consumption:
gases: 1-10 sccm/min
liquids: 1-5 μ l/min

Remaining work plan for FY 2016:

- Reduce temperature measurement uncertainty
- Continue and improve measurements for liquid reference fuels (iso-octane, n-heptane & toluene)
- Redesign burner to prepare for experiments at elevated pressures



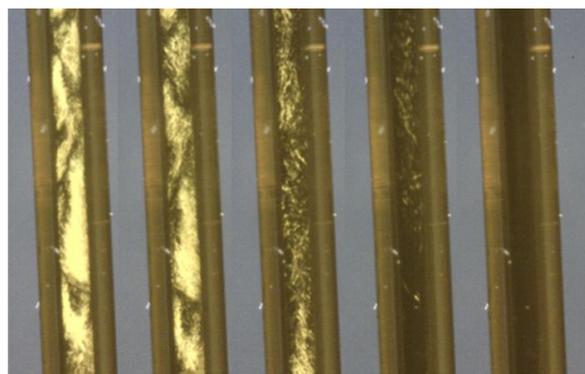


PNNL (Bays) 150k: Fuel Structure-Property Correlations

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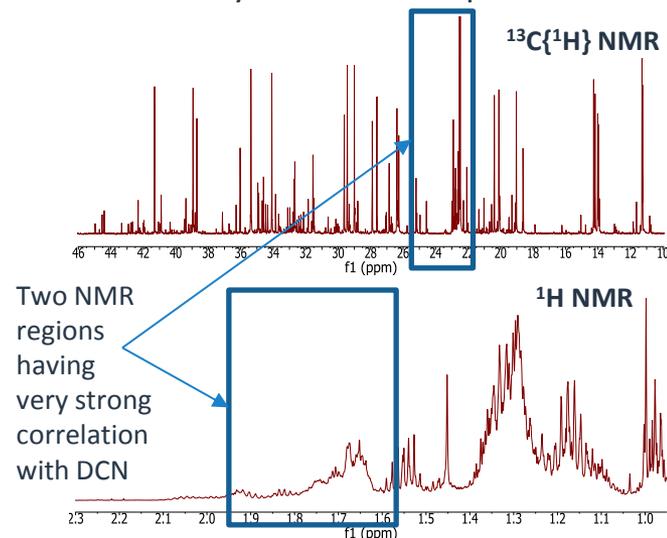
- Objective – Correlate fuel performance properties to chemical & solution structures in complex fuels
- Impact – Achieve the ability to *rapidly predict properties, like DCN, from < 1 mL of fuel*
- Approach
 - Use NMR/2D-GC to investigate fuel structure-property correlations, emphasizing DCN/RON prediction
 - Understand high-pressure phase behavior of fuels
- Future Work
 - Correlations to other properties, e.g., distillation or P_{vap}
 - P-T behavior beyond diesel surrogate fuels to fuels relevant to Co-Optima

Final Melting Point of *n*-Octadecane Crystals in Surrogate 3 at 10,000 psi*

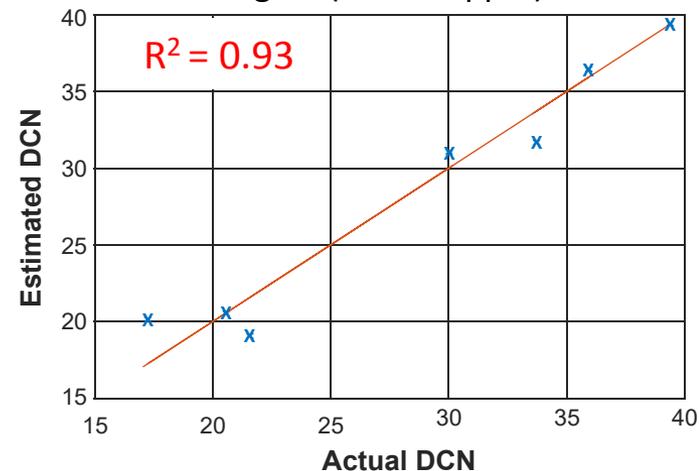


29 °C (Crystals) → 30 °C → 31 °C → 32 °C → 33 °C (Liquid)

^{13}C and ^1H NMRs of an Upgraded Biomass Hydrothermal Liquefaction Oil



DCN Prediction from a *Single* ^{13}C NMR Region (24.5-22 ppm)



*Results are preliminary. Final results are expected to be within 10% of presented values. Work conducted in collaboration with CRC Project AVFL-18a.

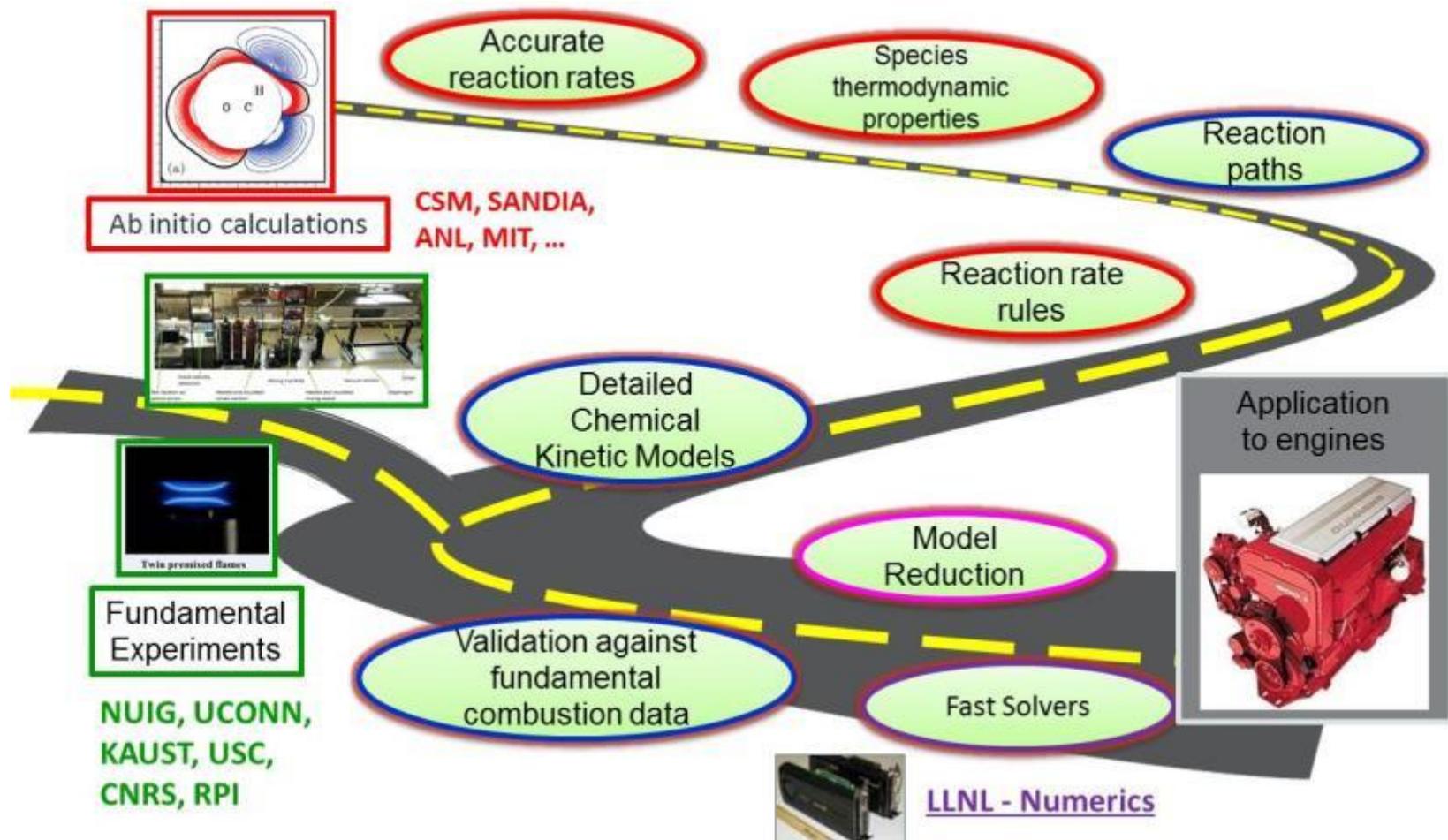


LLNL (Pitz) \$500k: Kinetic Mechanism Development

Objective:

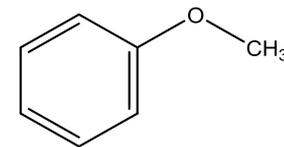
- Develop and validate chemical kinetic models for various bio-blendstocks and fuel surrogates

Approach:



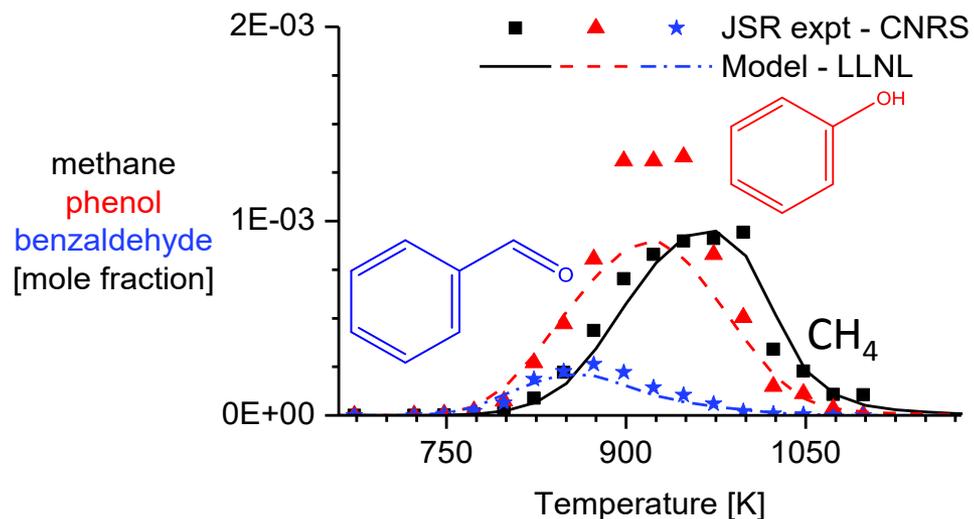
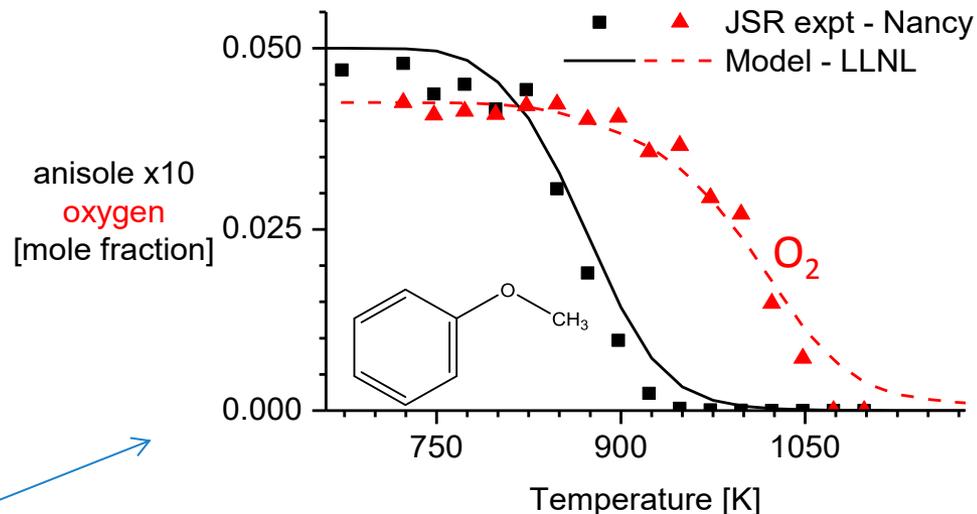


Pitz: Development of kinetic mechanism for anisole, a bio-derived fuel



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- Surrogate for methylated phenolics stream produced from biomass
- On Co-Optima list of most promising Thrust I candidates
- RON = 119, Sensitivity = 21
- LLNL preliminary mechanism complete
- Comparison of mechanism with intermediate species data
- Collaborators measuring ignition delay times (NREL), laminar burning velocities (Lund Univ.) and more intermediate species at high pressure (CNRS, Orleans) for additional validation



Jet stirred reactor experimental data:

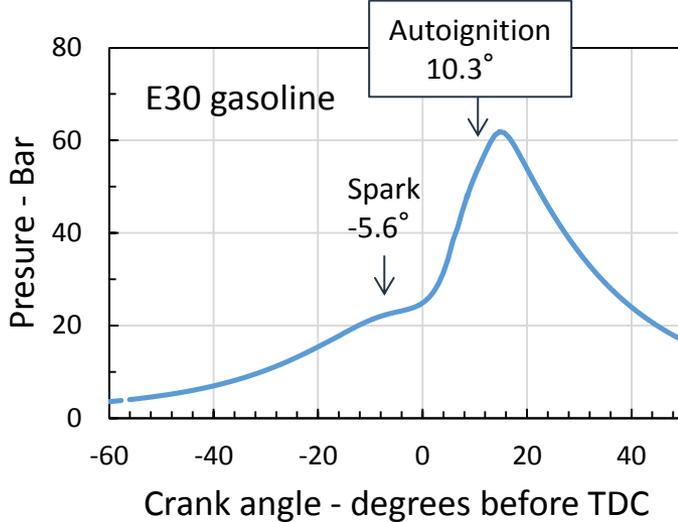
Nowakowska, Herbinet, Dufour, Glaude, Combust. Flame (2014)



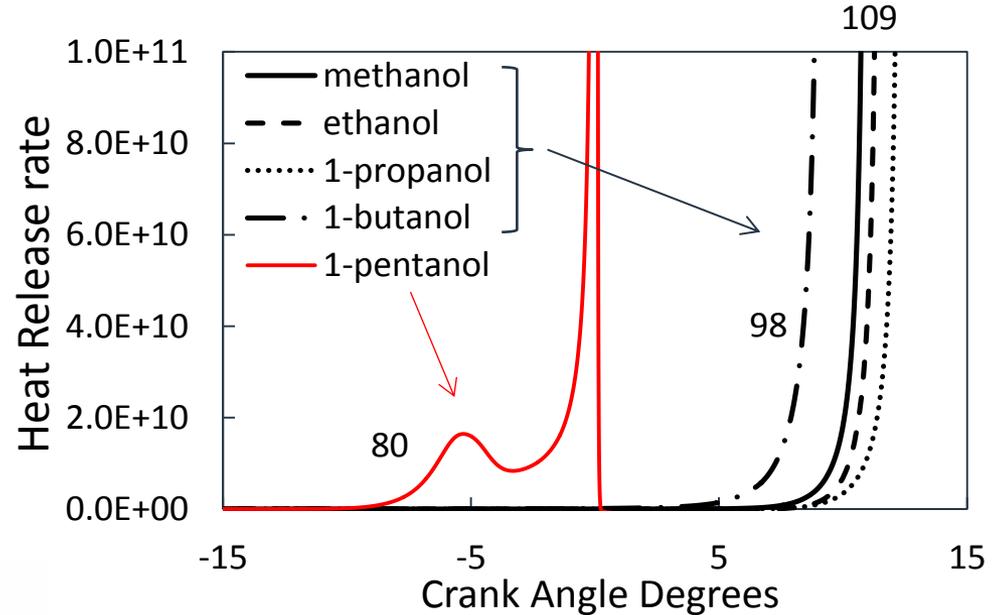
Pitz: Investigation of the effect of molecular structure on ignition behavior at RON-like engine conditions

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- RON-like pressure history from Magnus Sjöberg's DISI engine at Sandia

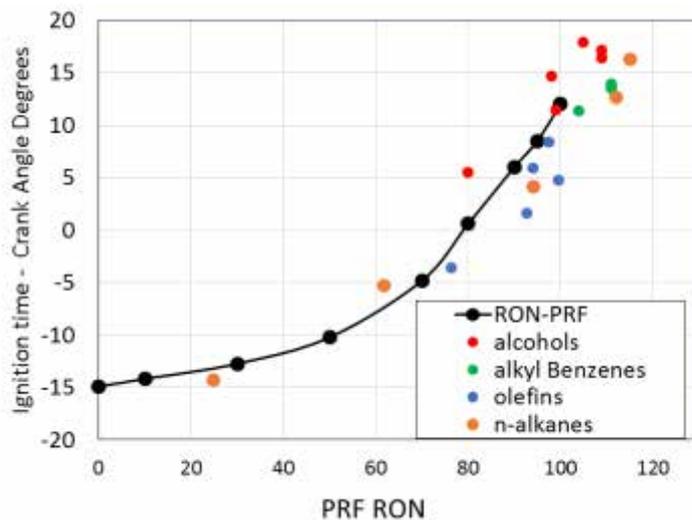


- Heat release curves for different alcohol chain-lengths for a RON-like engine pressure trajectory:



(Numbers on plot are RON values)

- RON predictions for a series of fuels in different chemical classes using RON-like pressure history

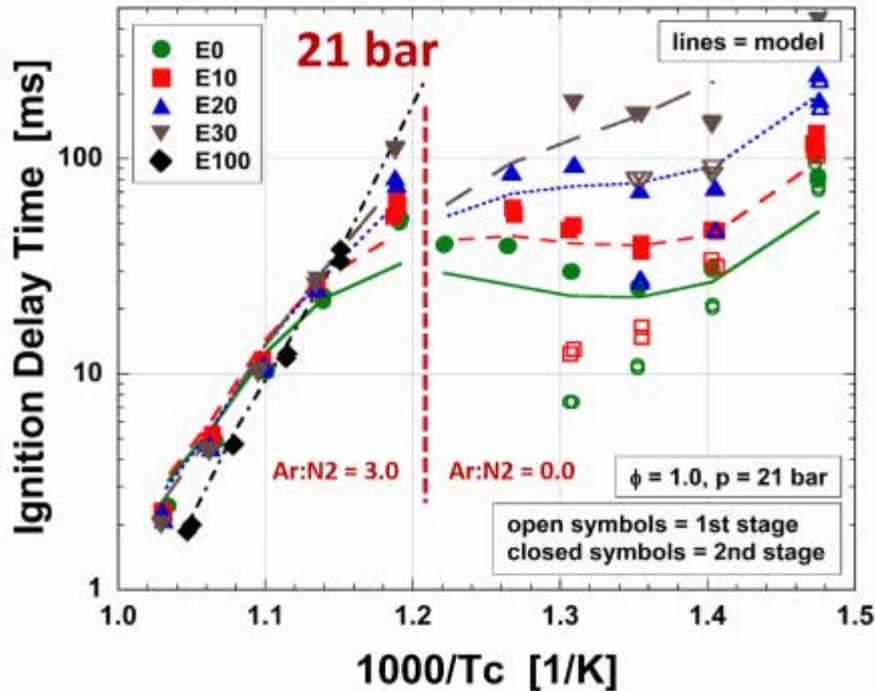




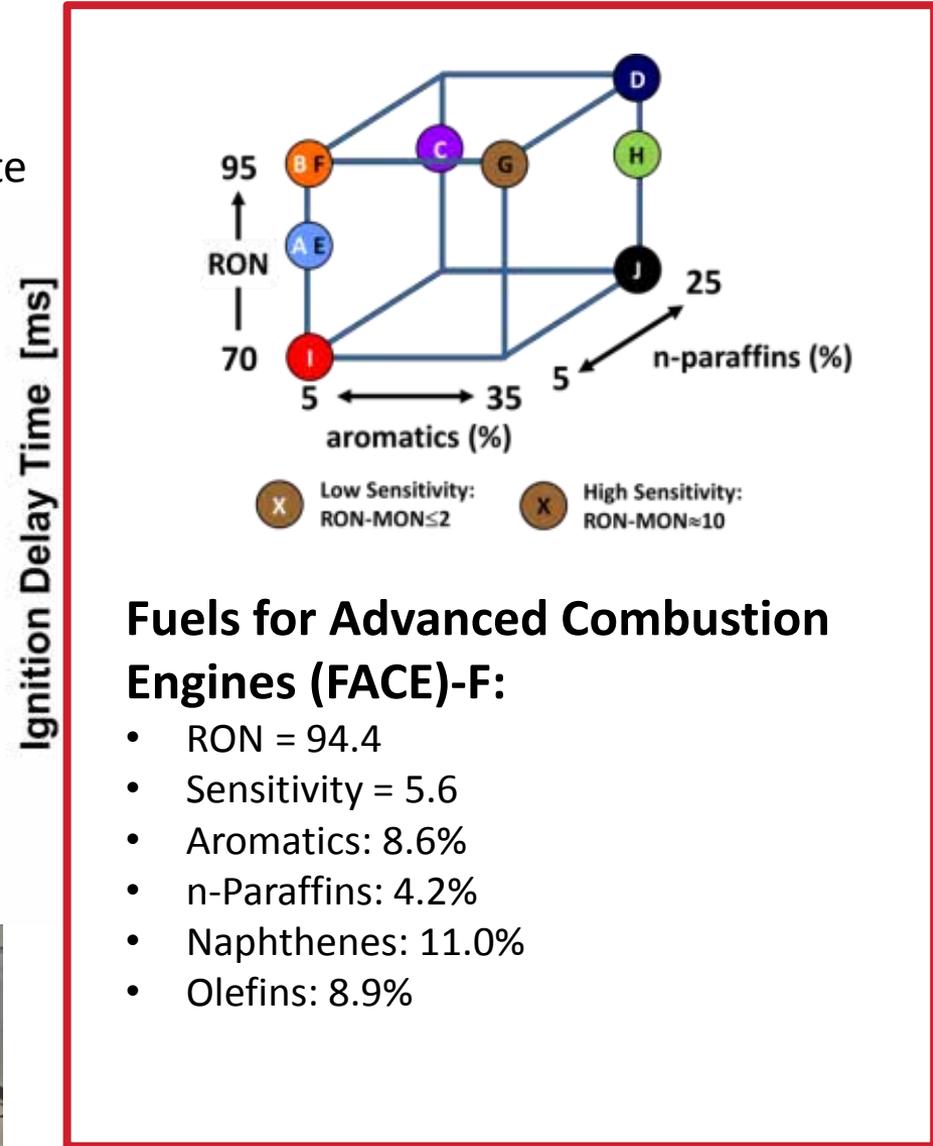
ANL (Goldsborough) \$250k: Rapid compression machine (RCM) investigation of gasoline mixtures with ethanol and validation of surrogate mechanism

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- ANL rapid compression machine
- CRC FACE-F / Ethanol blends (E0–E30, E100)
- LLNL kinetic mechanism for gasoline surrogate



Ignition Delay Time [ms]



Fuels for Advanced Combustion Engines (FACE)-F:

- RON = 94.4
- Sensitivity = 5.6
- Aromatics: 8.6%
- n-Paraffins: 4.2%
- Naphthenes: 11.0%
- Olefins: 8.9%

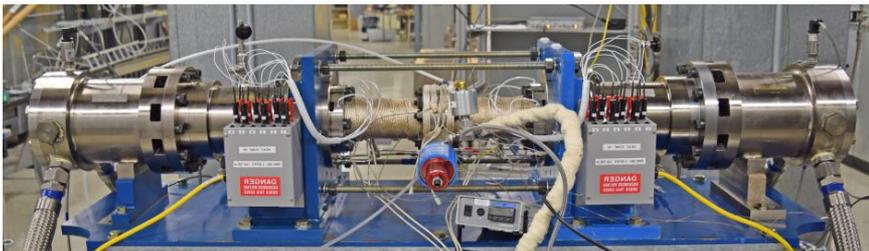
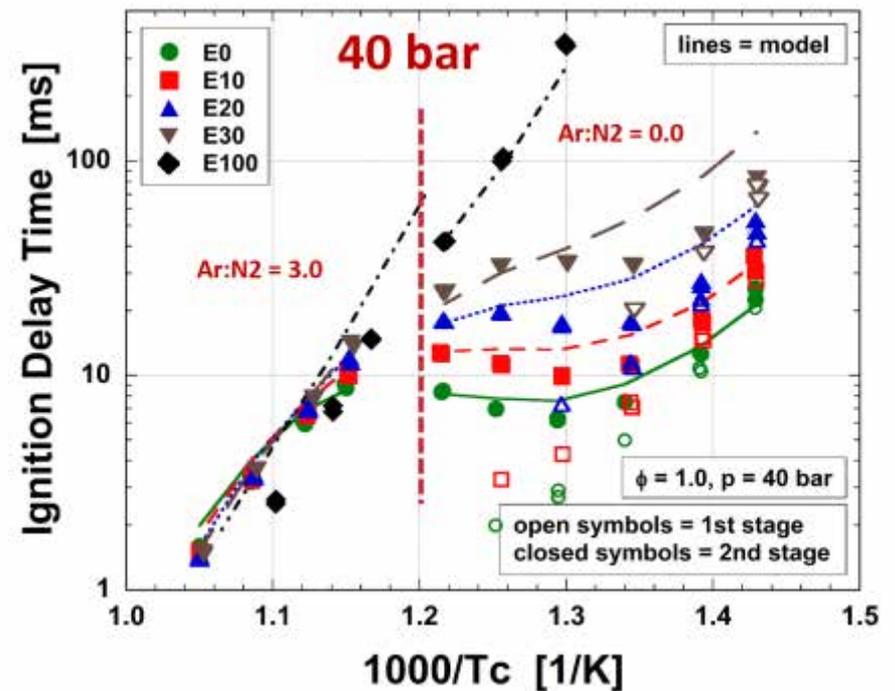
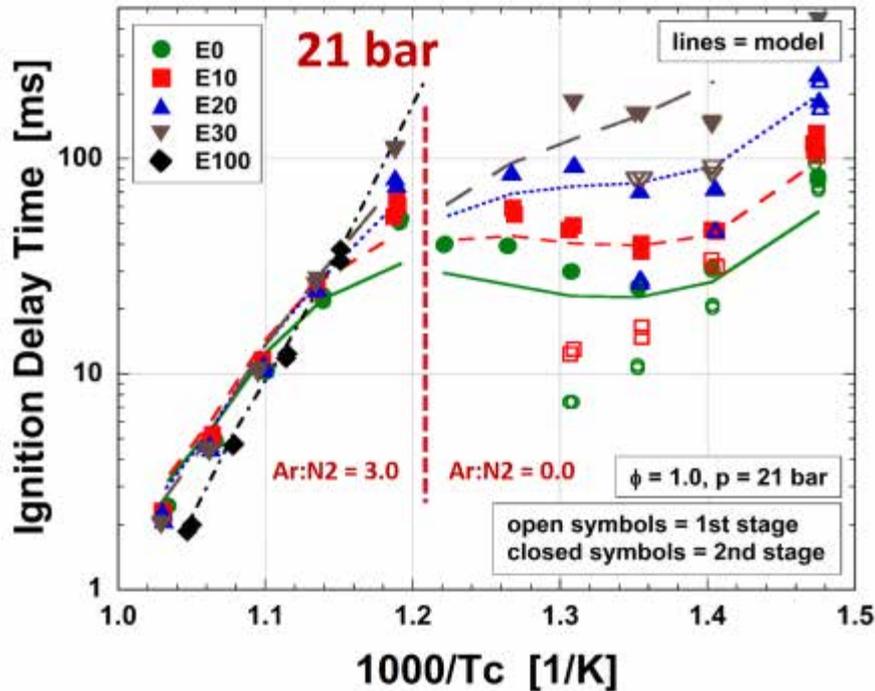




ANL (Goldsborough) \$250k: Rapid compression machine (RCM) investigation of gasoline mixtures with ethanol and validation of surrogate mechanism

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- ANL rapid compression machine
- CRC FACE-F / Ethanol blends (E0–E30, E100)
- LLNL kinetic mechanism for gasoline surrogate





NREL (Zigler) \$550K: Chemical Kinetics and SI Autoignition Behavior

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Objectives

- Assist in development and validation of chemical kinetic mechanisms for blends through ignition delay experiments and simulations.
- Combine bench-scale autoignition studies with engine experiments to more extensively quantify fuel blend ignition performance than possible with RON and MON.

Approach

- Employ a modified Ignition Quality Tester (IQT) to measure ignition performance of biofuel compounds blended into both key simple surrogate mixtures, and more complex gasoline range research fuels.
- Perform computational fluid dynamics (CFD) simulations of IQT to evaluate reduced kinetic mechanisms for blends against experimental data.
- Evaluate IQT-based data to highlight key kinetic behavior related to how ignition delay increases at low temperatures in relation to increased octane sensitivity.
- Characterize all fuels used in NREL's engine experiments (and several engine studies from other labs, including gasoline compression ignition) with the IQT.
- Begin correlation of IQT data to fuel chemistry and engine results.

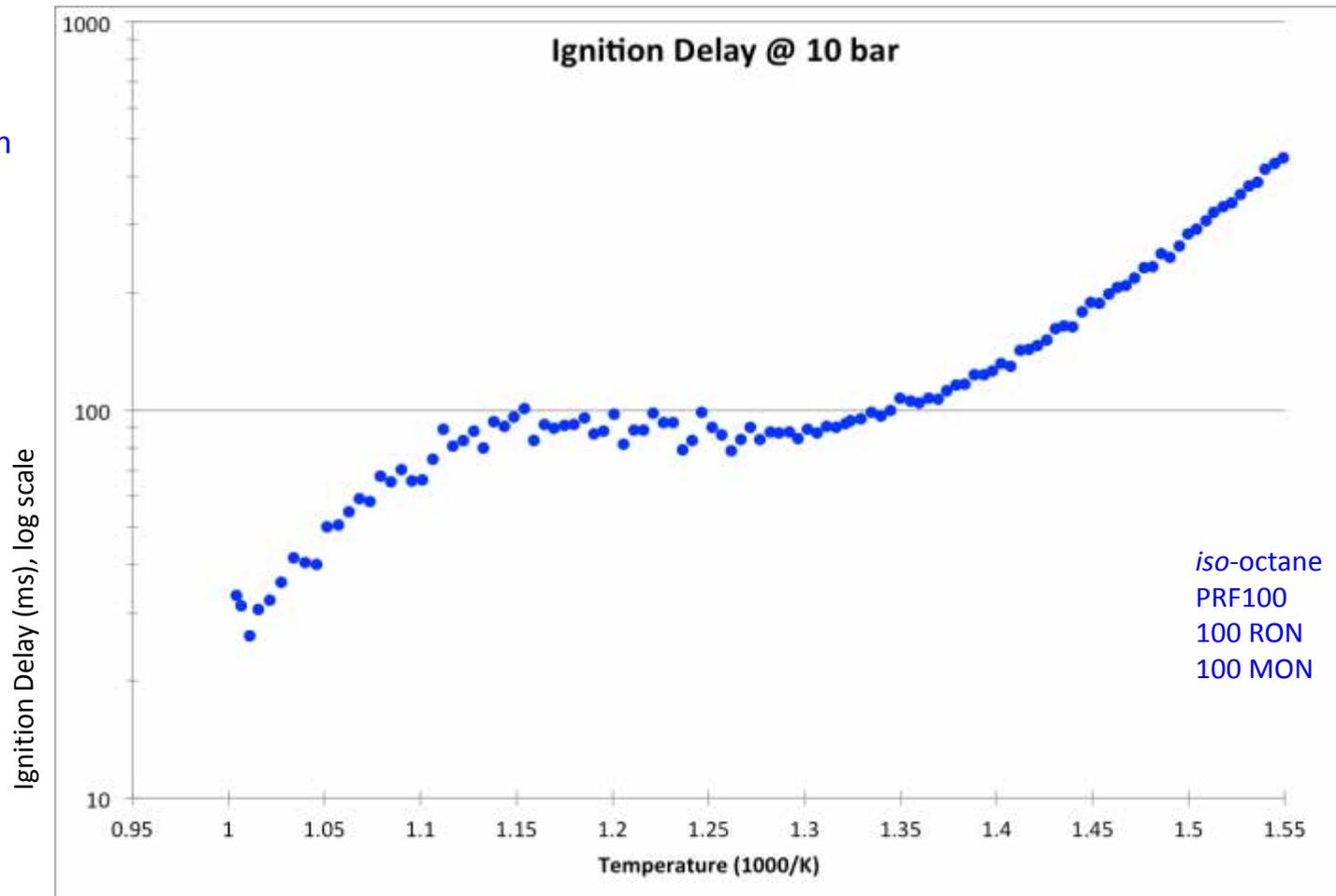


Zigler: Ignition delay experiments assist in development of kinetic mechanisms for biofuel blends

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- Temperature sweeps at several fixed pressures provide experimental data to develop accurate *kinetic mechanisms for blends* of biofuels into key gasoline surrogates.
- Simulations of IQT enable a development feedback loop for validation of mechanisms against engine-relevant experimental data.

- The critical negative temperature coefficient (NTC) region is mapped.



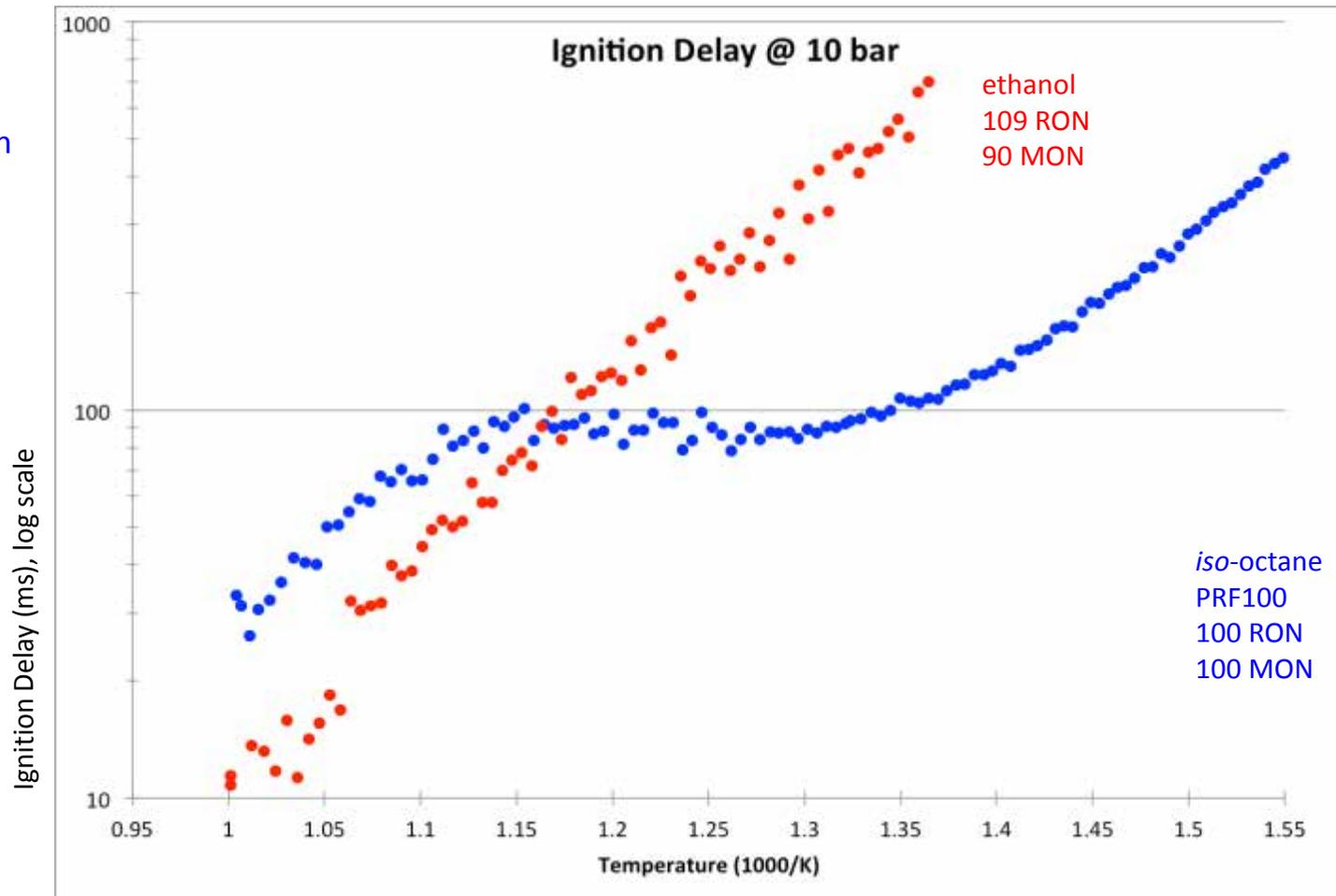


Zigler: Ignition delay experiments assist in development of kinetic mechanisms for biofuel blends

24

- Temperature sweeps at several fixed pressures provide experimental data to develop accurate *kinetic mechanisms for blends* of biofuels into key gasoline surrogates.
- Simulations of IQT enable a development feedback loop for validation of mechanisms against engine-relevant experimental data.

- The critical negative temperature coefficient (NTC) region is mapped.
- Note ethanol (109 RON) has shorter ignition delay at high T than *iso*-octane (100 RON).



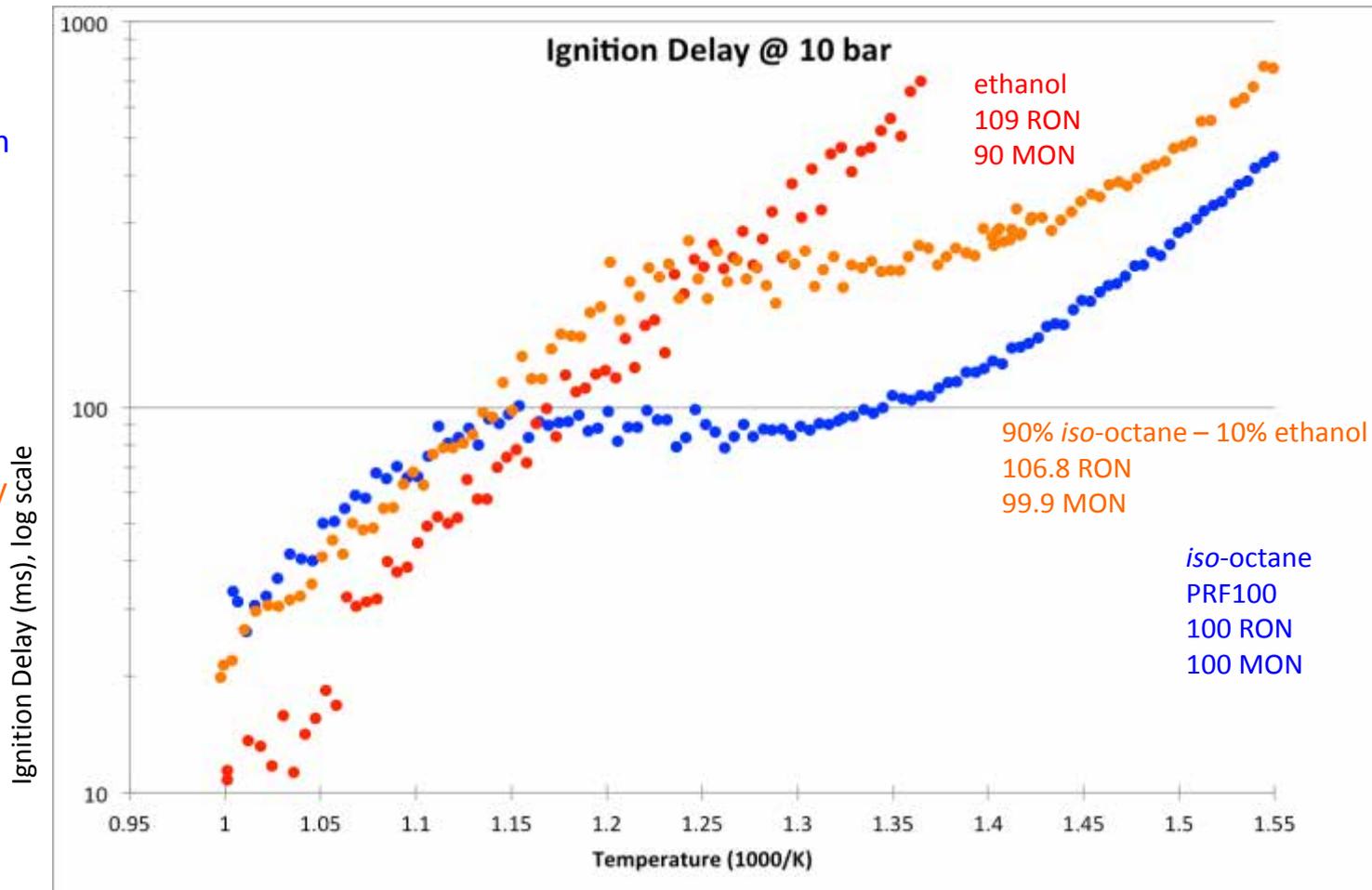


Zigler: Ignition delay experiments assist in development of kinetic mechanisms for biofuel blends

25

- Temperature sweeps at several fixed pressures provide experimental data to develop accurate *kinetic mechanisms for blends* of biofuels into key gasoline surrogates.
- Simulations of IQT enable a development feedback loop for validation of mechanisms against engine-relevant experimental data.

- The critical negative temperature coefficient (NTC) region is mapped.
- Note ethanol (109 RON) has shorter ignition delay at high T than *iso*-octane (100 RON).
- A 10% ethanol blend increases ignition delay while maintaining NTC behavior,



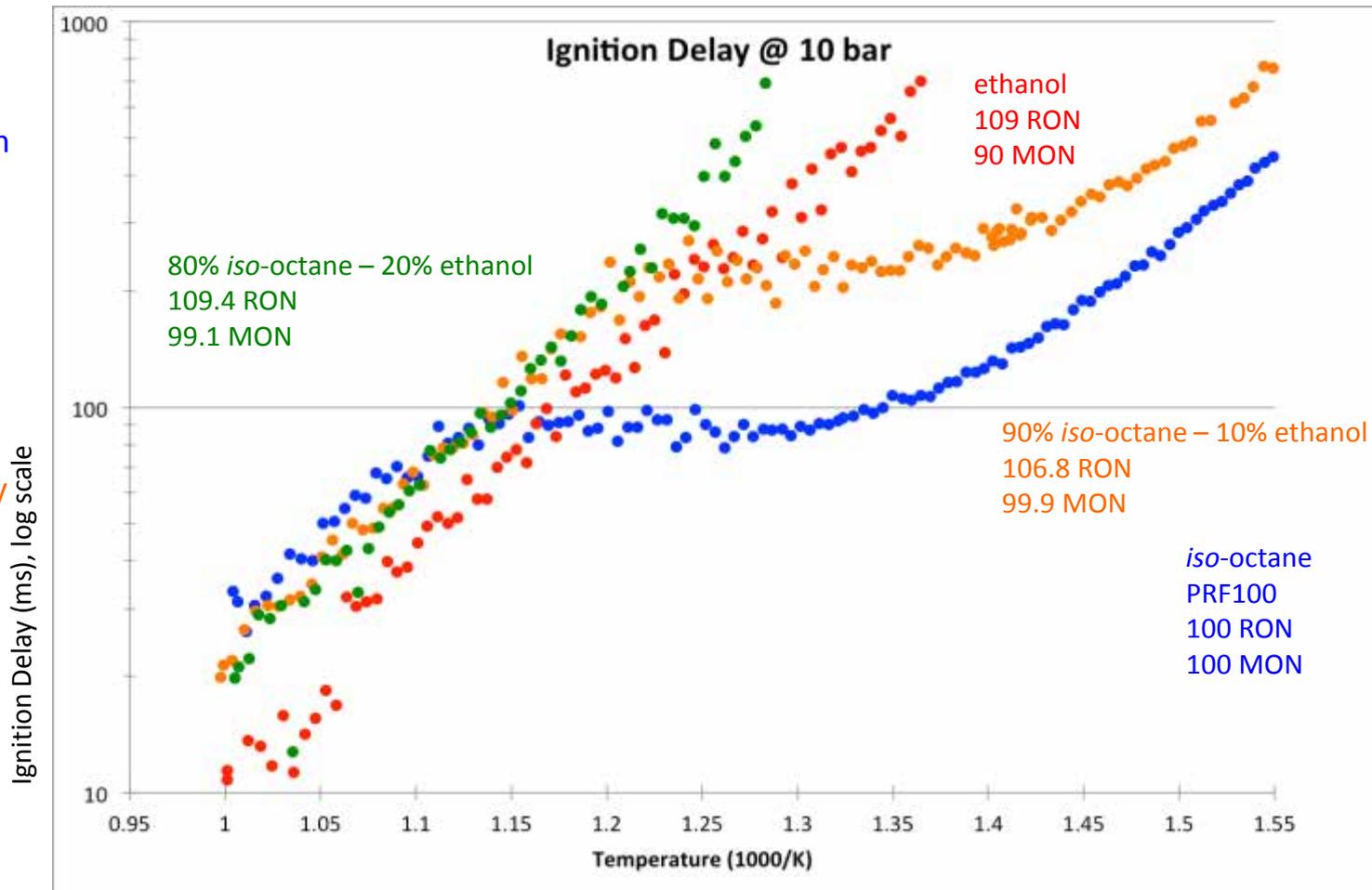


Ignition delay experiments assist in development of kinetic mechanisms for biofuel blends

26

- Temperature sweeps at several fixed pressures provide experimental data to develop accurate *kinetic mechanisms for blends* of biofuels into key gasoline surrogates.
- Simulations of IQT enable a development feedback loop for validation of mechanisms against engine-relevant experimental data.

- The critical negative temperature coefficient (NTC) region is mapped.
- Note ethanol (109 RON) has shorter ignition delay at high T than *iso*-octane (100 RON).
- A 10% ethanol blend increases ignition delay while maintaining NTC behavior,
- But a 20% ethanol blend eliminates NTC behavior while increasing ignition delay over neat ethanol in that region.

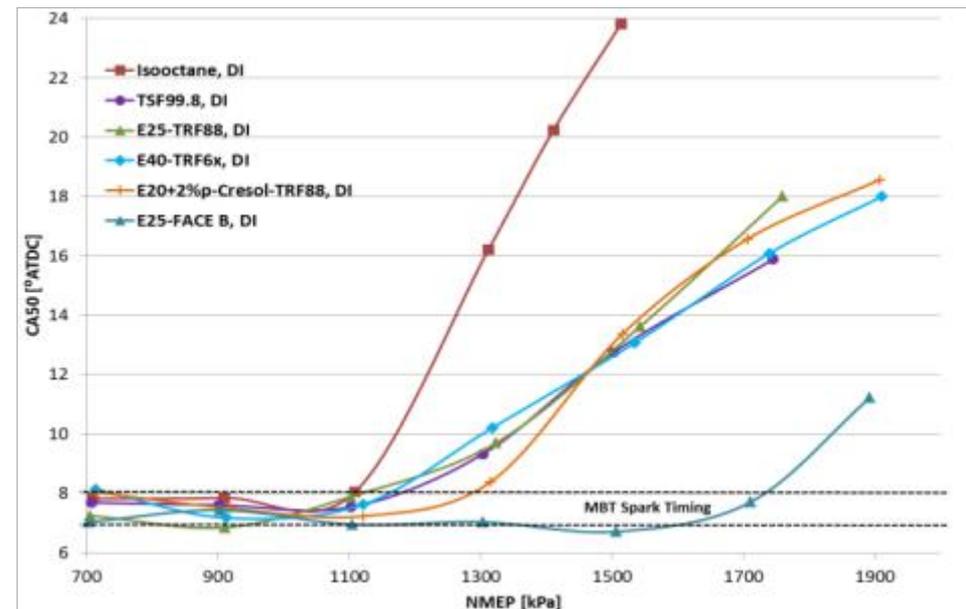
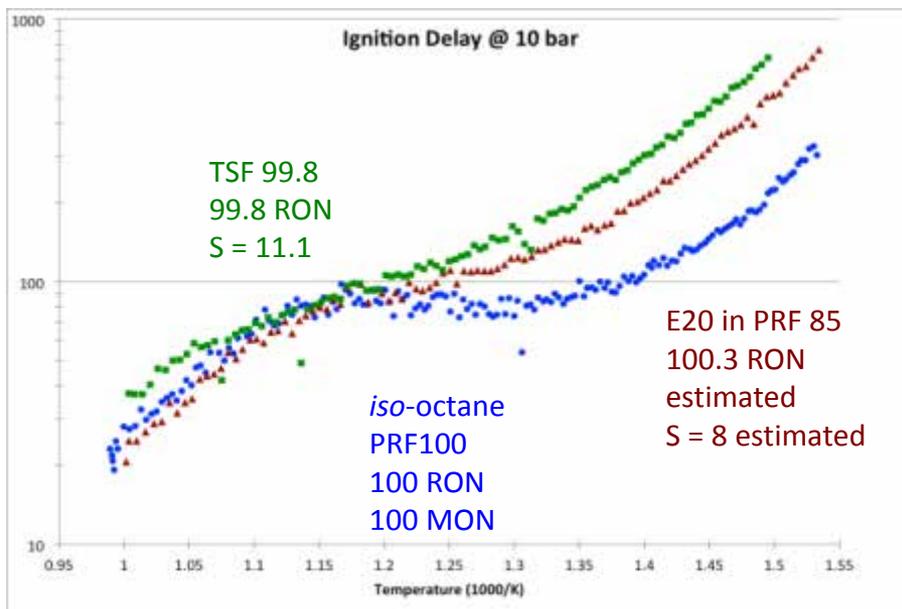




Zigler: Ignition delay experiments provide more insight to engine results beyond RON and octane sensitivity

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- Temperature sweeps at several fixed pressures provide experimental ignition delay data (rather than just RON or MON), including *how ignition delay increases at low temperatures in relation to increased octane sensitivity (S)*.
 - The IQT data *will be correlated with engine data* focusing on load extension possible using spark retard with high S fuels.
-
- In this study of ~100 RON fuels, increasing octane sensitivity correlates with increased ignition delay at lower temperatures.
 - IQT data offer insight why increased load is possible with retarded spark timing, where end-gas follows lower temperature trajectory with increased ignition delay times.
 - Engine studies show that higher S enables increased load via spark retard (following presentation).



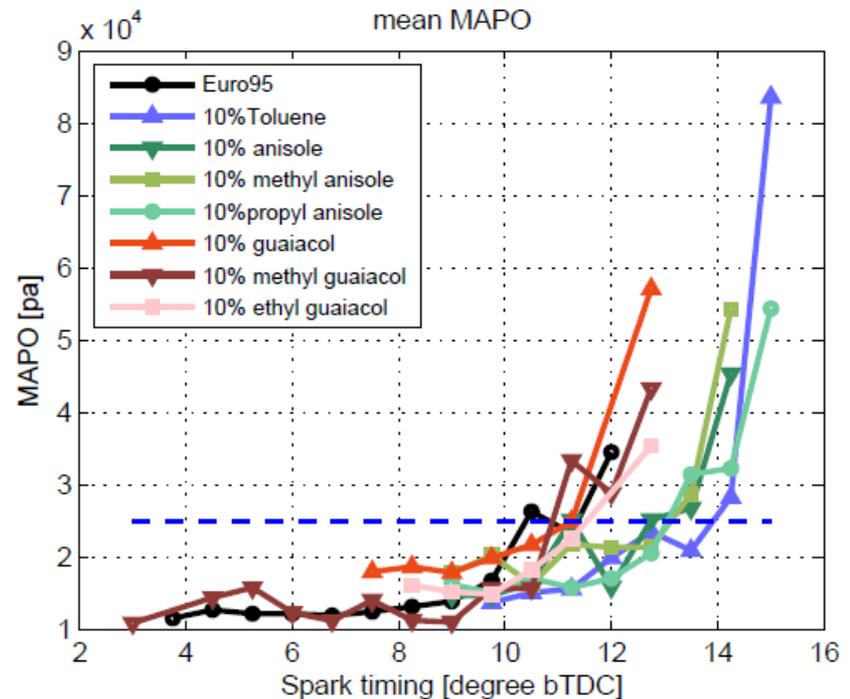
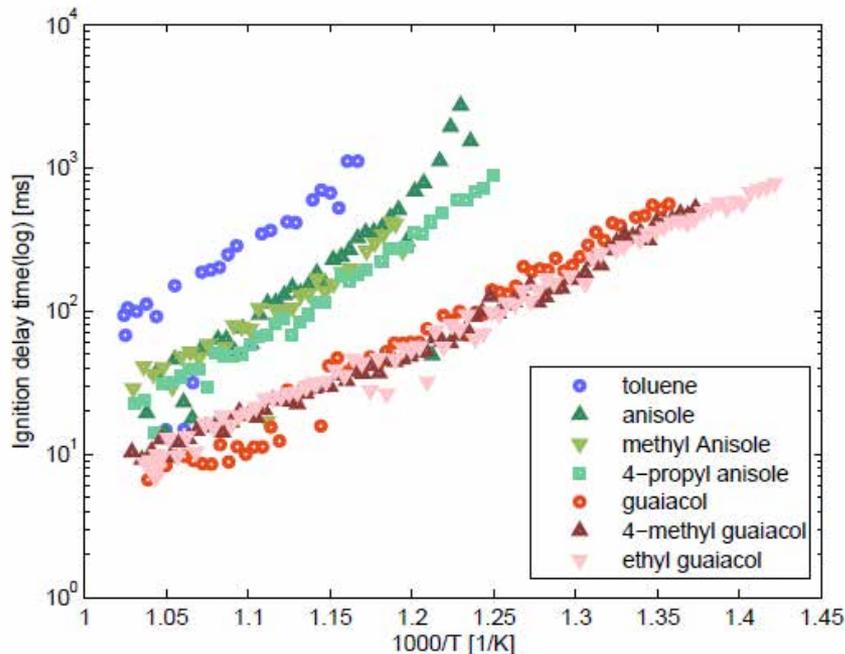


NREL: Ignition delay experiments assist in development of kinetic mechanisms for biofuel blends

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- Temperature sweeps at several fixed pressures provide experimental ignition delay data (rather than just RON or MON), to examine biomass-oxygenate structure effects on engine knock resistance.
- IQT data is being correlated with engine data to explain oxygenate structure effects.

Tian, M., Ratcliff, M., McCormick, R.L., Luecke, J., Yanowitz, J., Glaude, P.-A., Cuijpers, M., Boot, M.D., “Evaluation of Lignin Based Octane Boosters in a Spark Ignition Engine” submitted.





Future work: kinetics

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Pitz

- Develop surrogate mixture models for gasoline fuels to be used in Co-Optima:
 - Three RON 98 reference gasoline fuels: alkylate, reformat and ethanol-containing
 - Base fuel(s) selected for blending with bio-derived blendstocks
- Simulate SI and CI engine experiments :
 - with RON 98 gasolines for
 - constant octane sensitivity obtained from aromatics or ethanol
 - octane sensitivity varied
 - using basefuel(s) combined with Tier 2 bio-derived blendstocks
- Use kinetic models to investigate high-sensitivity fuels deriving octane sensitivity from aromatics or olefins. Is there a significant difference autoignition resistance at Thrust 1 conditions?
- Develop/validate component kinetic models to represent missing chemical classes in Tier 2 blendstocks (ketones and methyl furan)
- Investigate blendstocks that can probe the most beneficial fuel properties (resistance to autoignition and high flame speed) for Thrust 1 conditions.

Zigler

- Expand IQT ignition kinetics studies of key surrogate blends beyond ethanol to include other biofuel compounds of interest, in coordination with kinetics sub-team (within FP team), and AED and LGGF teams
- Incorporate IQT-based parametric ignition delay data in engine simulation knock integral calculations to correlate to engine-based experimental data
- Focus more experimental studies and kinetic model simulations on relationship between octane sensitivity, increased ignition delay time at lower temperatures, and engine load extension possible with spark retard and high S (includes tighter collaboration with LLNL)
- Expand collaboration with other labs on studying fuels used in their engine experiments in the IQT (e.g., collaboration with ANL on gasoline compression ignition)
- Begin industry collaboration (including via CRC) of using IQT-based experimental studies to provide more ignition performance data for fuel / engine studies



Fuel Properties Team Research Summary

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Relevance

Fuel Properties research is the crucial link between fuel production (LGHG Fuels Team) and engine combustion (Advanced Engine Development Team) and testing of the central fuel properties hypothesis investigated within “Co-Optima.”

Approach

Careful measurement of fuel properties, development of new fuel property measurement methods, and targeted engine experiments directed at revealing fuel property effects – all based on a fuel matrix designed to go well beyond the chemistry represented by conventional petroleum-derived fuels.

Accomplishments

- In collaboration with LGHG Fuels Team, a fuel property database was constructed and populated
- A three tier fuel screening process was developed. Tier 1 screening for Thrust I (SI) fuels was performed and a list of most promising advanced SI engine bio-blendstocks developed
- New methods for measuring fuel heat of vaporization were developed and are being refined
- Small volume autoignition testers are being developed for rapid fuel screening using milliliter volumes
- Correlations between easily measured chemical parameters and fuel properties are being developed
- Validated combustion kinetic models for important bio-blendstocks and fuel surrogates are being developed based on RCM and IQT kinetic data

Collaborations

- “Co-Optima” has 9 National Labs, stakeholder engagement, and external advisory board
- Projects presented also represent extensive collaborations with industry and universities

Future Work – A portfolio of ongoing and future work is described



Co-Optimization of Fuels and Engines Fuel Property and Advanced Engine Development Team

Thrust I engine projects

June 9th, 2016

Jim Szybist,¹ Scott Sluder,¹ Matt Ratcliff,² Thomas Wallner,³ Derek Splitter,¹ Andrew Ickes,³ Christopher P. Kolodziej,³ Bob McCormick,² Paul Miles⁴

1. Oak Ridge National Laboratory
2. National Renewable Energy Laboratory
3. Argonne National Laboratory
4. Sandia National Laboratories

Co-Optima DOE VTO Management Team: Kevin Stork and Gurpreet Singh



overview: thrust I engine projects

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Thrust I engine projects focus on Spark Ignition combustion strategies with a focus of understanding the effects of fuel properties

Overarching Fuel Property Hypothesis: If we understand the *critical fuel properties* correctly, then fuels with those properties will provide comparable performance regardless of the chemical composition.

Merit Function

Efficiency Benefits of High Octane Fuels

ORNL Sluder

Effects of RON, HoV, and Octane Sensitivity

NREL Ratcliff

ANL Kolodziej/Ickes

Dilution Limits on SI Combustion

ORNL Szybist

ANL Kolodziej/Wallner

Fuel Effects on LSPI

ORNL Splitter



milestones for Thrust I engine projects are either complete or on-track

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Tracked co-optima milestones

- Q2. Quantify vehicle fuel efficiency gains from the use of a ~98 RON gasoline in combination with increased compression ratio using the Ford 1.6L engine (ORNL Sluder) **Complete**
- Q3. Determine the strengths and deficiencies that exist in using flame speed to predict dilution tolerance through experimental and kinetic modelling studies (ANL Kolodziej/Wallner) **On-track**
- **Additional project level milestones are also on-track**



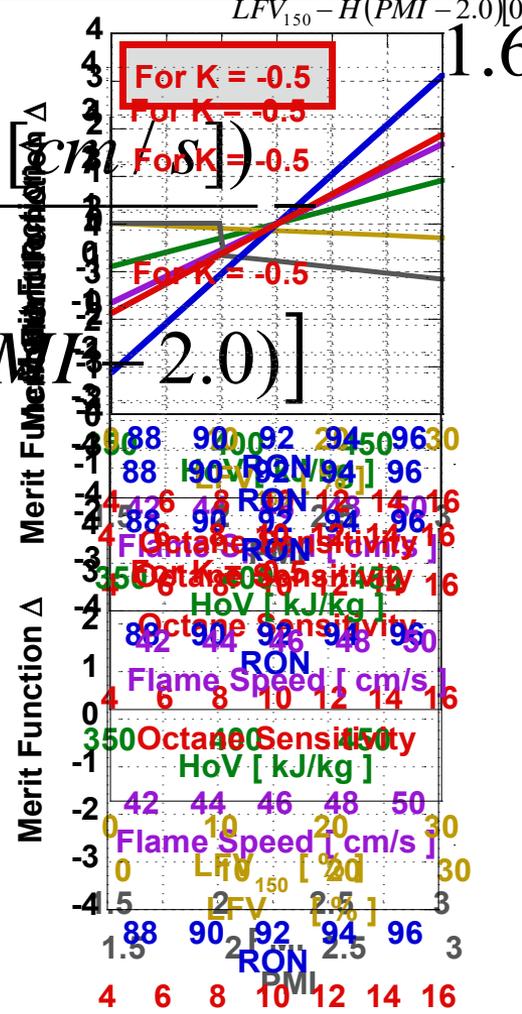
purpose of efficiency merit function for thrust I: tool to rank promising candidates for further study

$$\text{Merit}_{mix} = \frac{1}{1.6} \left[\frac{(\text{RON}_{mix} - 92)}{1.6} + \frac{0.01[\text{ON}/\text{kJ}/\text{kg}](\text{HoV}_{mix} - 415)[\text{kJ}/\text{kg}]}{1.6} + \frac{(\text{RON}_{mix} - 92)}{1.6} - K \frac{(S_{mix} - 10)}{1.6} \right] + \frac{1}{1.6} \left[\frac{(\text{RON}_{mix} - 92)}{1.6} + \frac{0.01[\text{ON}/\text{kJ}/\text{kg}](\text{HoV}_{mix} - 415)[\text{kJ}/\text{kg}]}{1.6} + \frac{(\text{RON}_{mix} - 92)}{1.6} - K \frac{(S_{mix} - 10)}{1.6} \right] + \frac{1}{1.6} \left[\frac{(\text{RON}_{mix} - 92)}{1.6} + \frac{0.01[\text{ON}/\text{kJ}/\text{kg}](\text{HoV}_{mix} - 415)[\text{kJ}/\text{kg}]}{1.6} + \frac{(\text{RON}_{mix} - 92)}{1.6} - K \frac{(S_{mix} - 10)}{1.6} \right] + \frac{1}{1.6} \left[\frac{(\text{RON}_{mix} - 92)}{1.6} + \frac{0.01[\text{ON}/\text{kJ}/\text{kg}](\text{HoV}_{mix} - 415)[\text{kJ}/\text{kg}]}{1.6} + \frac{(\text{RON}_{mix} - 92)}{1.6} - K \frac{(S_{mix} - 10)}{1.6} \right]$$

The goal of Thrust 1 engine research is to develop a robust and quantitative understanding on how efficiency is impacted by fuel properties

$$\text{LFV}_{150} - H(\text{PMI} - 2.0)[0.67 + 0.5(\text{PMI} - 2.0)]$$

- This is an **approach** on how to value properties for efficiency
- This is not a finished product, the merit function will continuously evolve
- We are working to determine if these are the right fuel properties whether we adequately understand their impacts
- The fuel property hypothesis will be tested with biofuels that introduce different chemistry (structures and functional groups)





ORNL (Sluder) \$500k: efficiency benefits of high octane fuels

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ORNL - Sluder (1/3)

Objectives

- Quantify fuel efficiency impacts of octane number
- Test central fuel property hypothesis using fuels with different chemistry

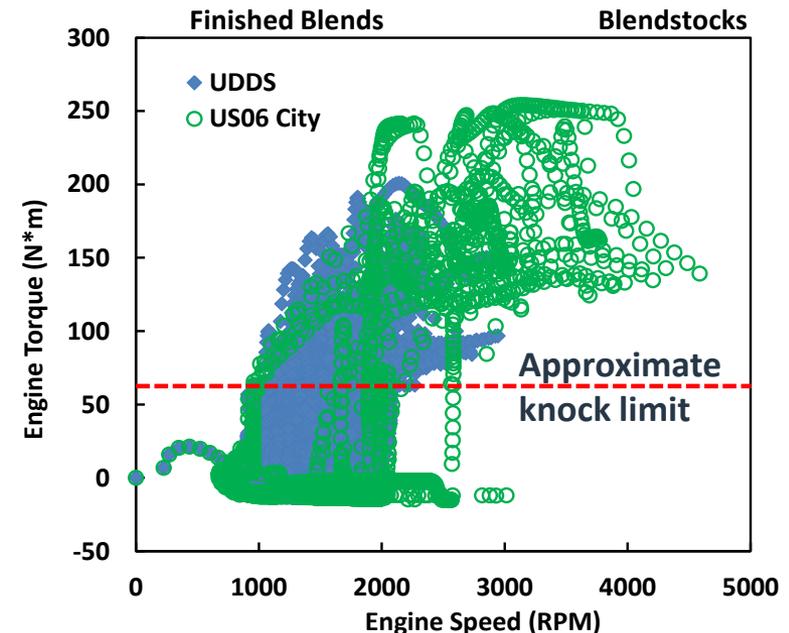
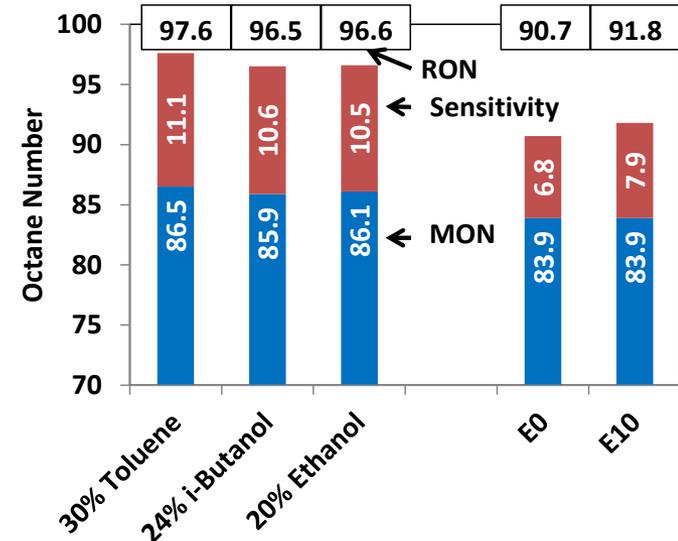
Approach

Engine Experiments

- Experiments with a 1.6L Ecoboost engine using three different compression ratios (10:1 (stock), **12:1**, and 13:1)
- Map engine performance with multiple fuel formulations

Vehicle Simulations

- Experimental data used in Autonomie vehicle simulations to quantify vehicle energy consumption and fuel economy



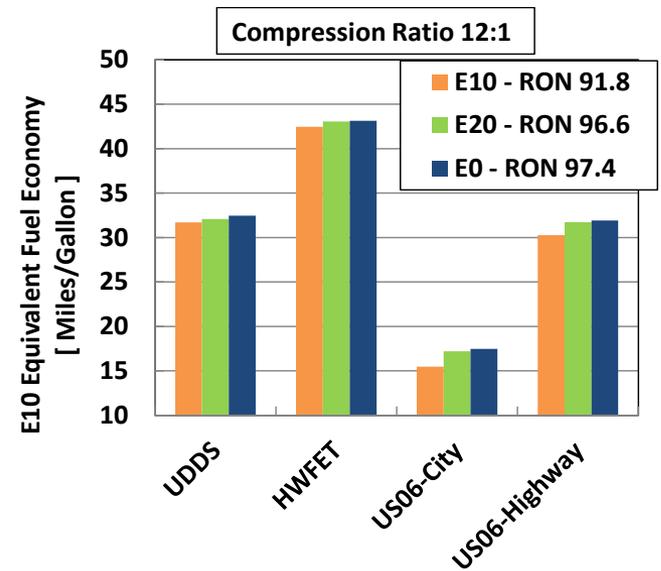
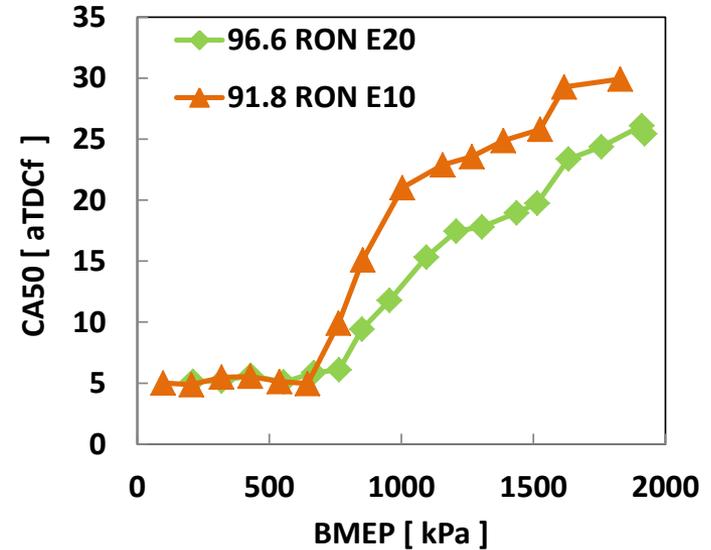


at fixed compression ratio, the benefit of increased octane rating for high-torque cycles can be significant

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ORNL - Sluder (2/3)

- Increasing octane rating reduces spark retard for knock avoidance
 - Improvement is limited to knock-limited load range
- Important to match RON to CR with desired torque
 - Downsizing/downspeeding trend demands increasing torque
- Energy consumption improvement for 5 RON points
 - UDDS cycle (light load): 2%
 - US06 city portion (higher load): 11%
- 3.5% energy consumption decrease needed to achieve volumetric fuel economy parity for E20 compared to E10



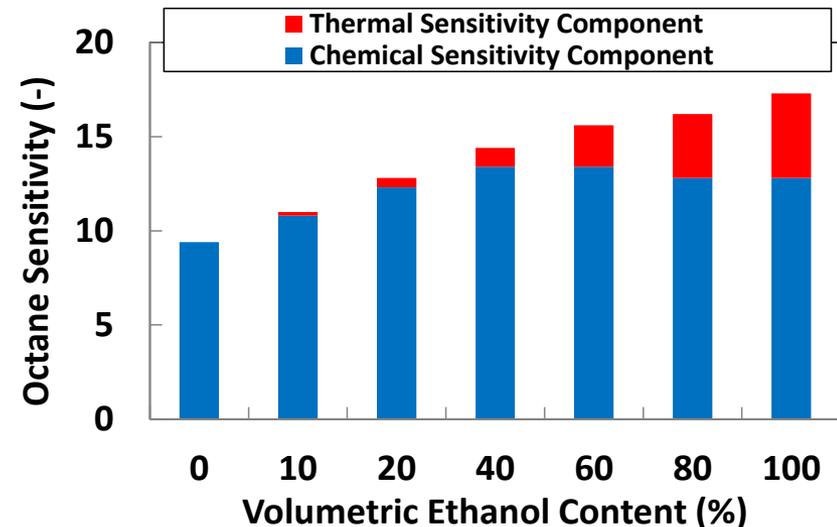
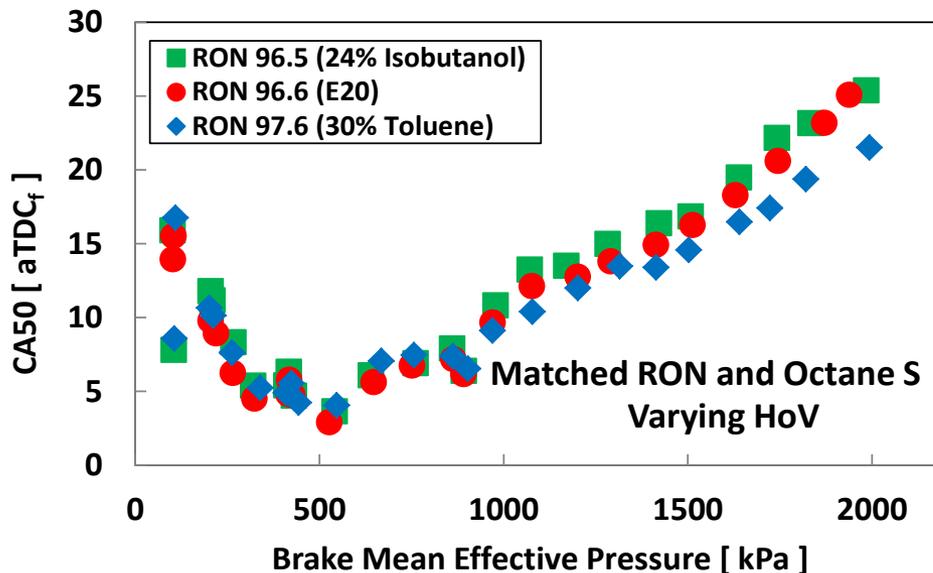
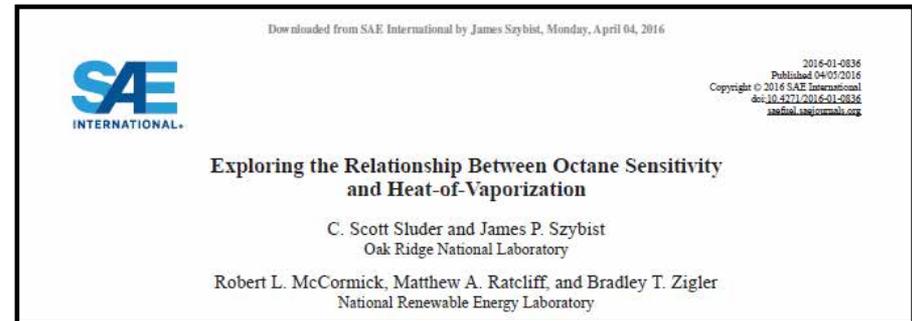


ORNL and NREL collaboration: HoV can be thought of a contributor to octane sensitivity

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ORNL - Sluder (3/3)

- Inconsistencies in literature regarding HoV impact on knock propensity
 - HoV effect only been observed when covariant with octane sensitivity
- Expanded with new experimental results from ORNL and NREL
- Main conclusion: HoV is a thermal contributor to sensitivity
 - Aligns the findings of seemingly contrary literature findings
 - Consistent with the vaporization effects in the RON and MON tests





NREL (Ratcliff) \$250k: effects of S, HoV and RON on GDI Performance

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NREL – Ratcliff (1/3)

Objectives

- Test null hypothesis: At a given octane sensitivity, HoV impact on knock resistance is included in S

Approach

- Single cylinder version of GM Ecotec 2.0L, 9.2: CR
- Side-mounted DI or upstream fuel injection
- Load sweeps at an intake manifold temp of 50 °C
- Sweep intake manifold T for max load at 2 different CA50 phasing

Fuel Matrix

- Matched RON and S at variety of HoV
- Fuels with different S and RON included to bound results

Fuel	RON	S	HOV [kJ/kg]	Oxygen wt%
Isooctane (PRF 100)	100	0	303	0
TSF99.8 (74% Toluene)	99.8	11.1	390	0
E25 in TRF88 (23% Toluene)	101.6	10.7	489	24.6
E40 in TRF6x (24% Toluene)	99.2	12.2	595	40.3
E20 + 2% p-Cresol in TRF88	100.3	10	472	22.4
E20 + 6% Anisole in TRF88	99.9	11	472	26.5
E25- FACE B	105.6	11.8	485	26.3

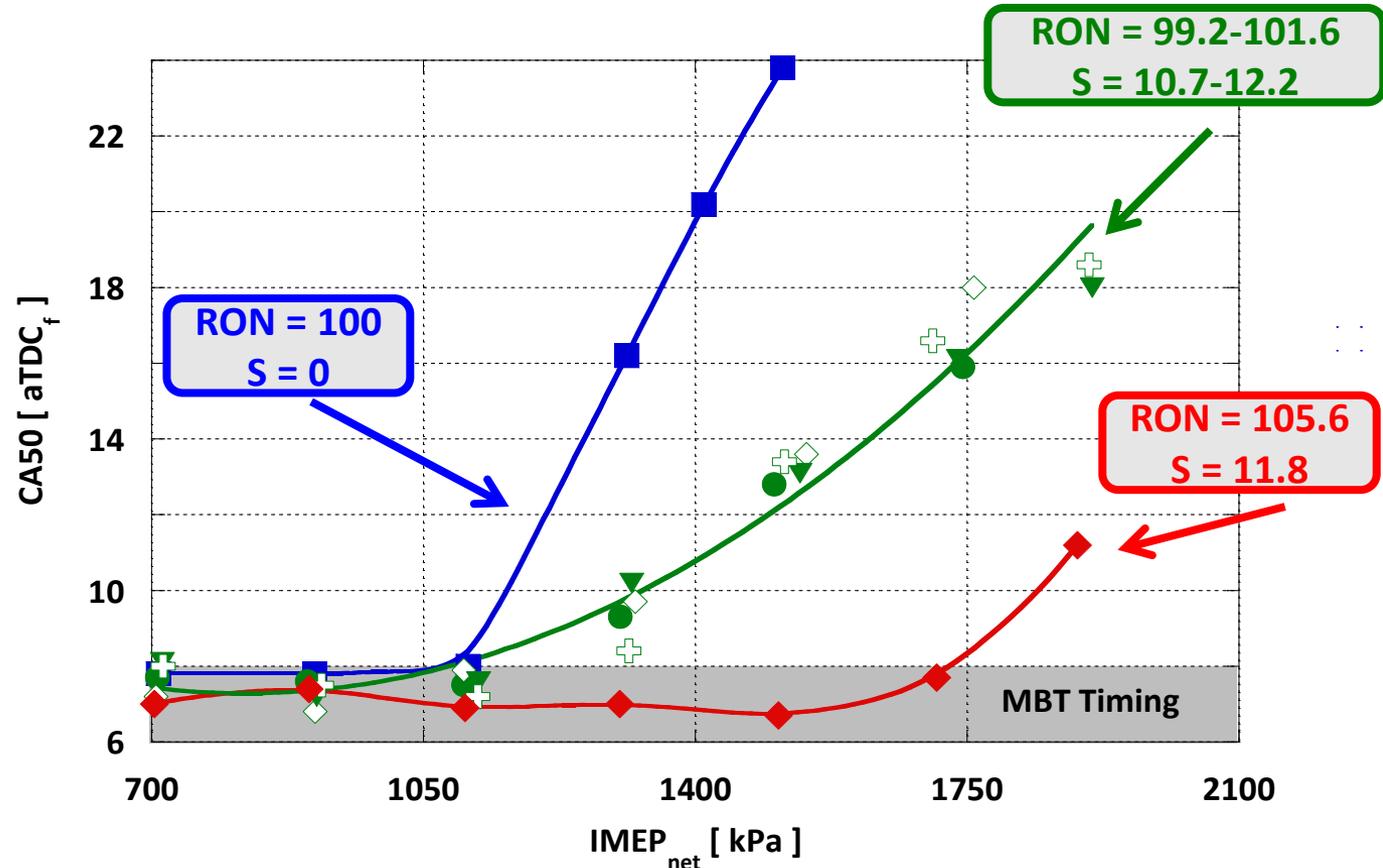
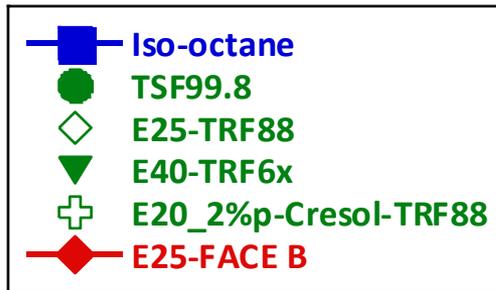


load sweeps reveal that performance of fuels with matched RON and S is independent of HoV

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NREL - Ratcliff (2/3)

- Using direct injection (DI) all 100 RON, $S \approx 11$ fuels have similar knock-limited performance gains over isooctane; no evident HoV benefit
- $S \approx 11$ allows more than half the combustion phasing advance available with 106 RON fuel, between 1200 – 1500 kPa NMEP





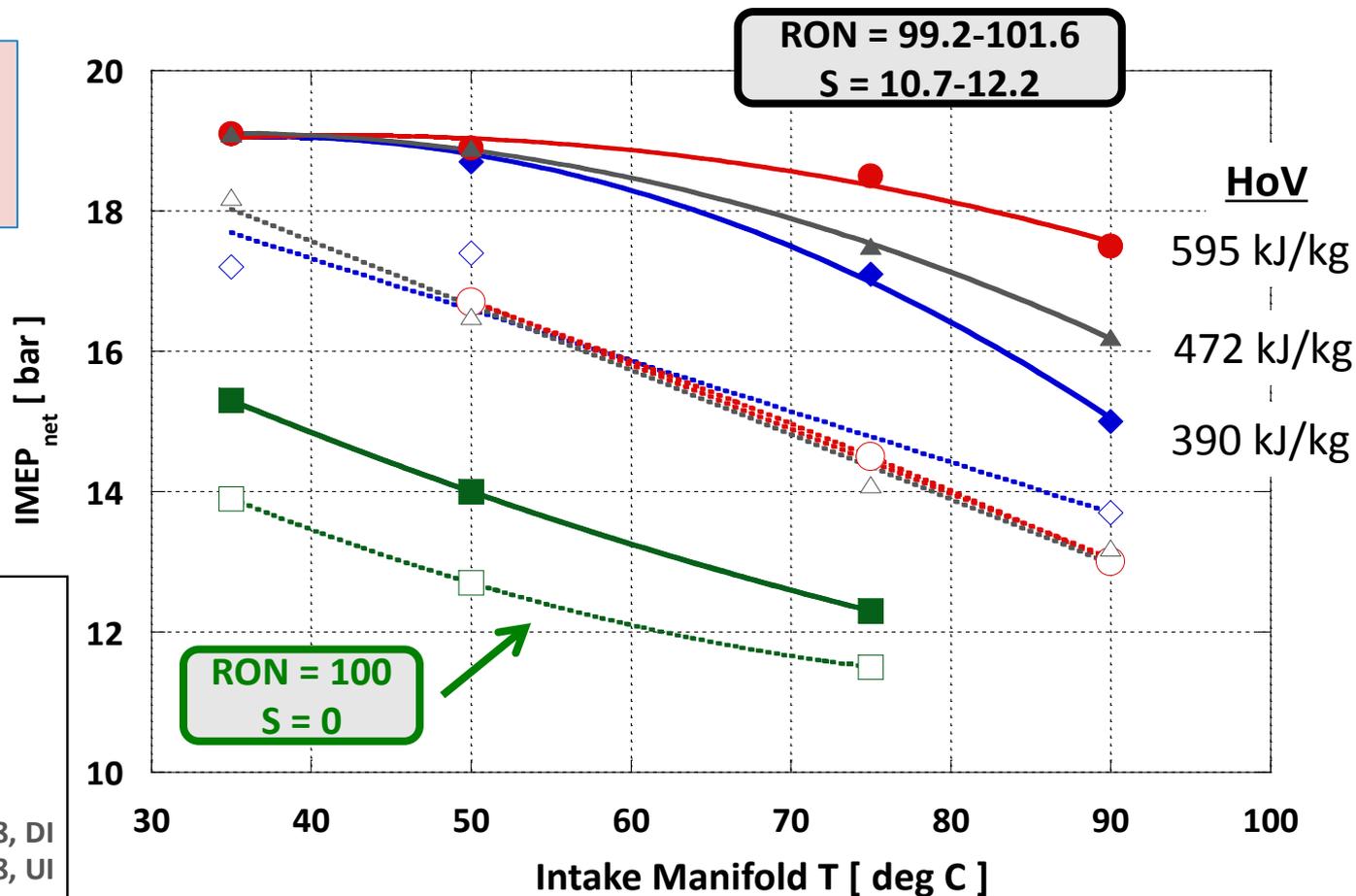
HoV appears to improve performance at elevated intake air temperatures

40

NREL - Ratcliff (3/3)

- Increasing HoV improves DI performance at late combustion phasing and intake manifold temperatures greater than 50 °C
- Performance of upstream injected (UI) 100 RON, S ≈ 11 fuels is significantly higher than DI of isooctane

Plot shows peak IMEP at a constant CA50 phasing





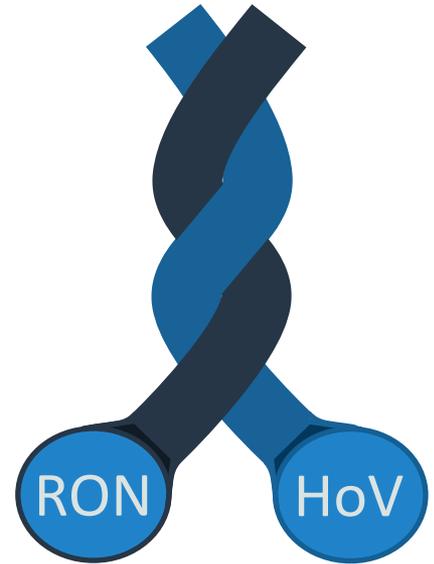
ANL (Wallner) \$300k: studies of research octane number (RON) and heat of vaporization (HoV)

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ANL– Kolodziej/Ickes (2/2)

- **Objectives**

- Develop a better understanding of how fuel properties, particularly RON and HoV, impact auto-ignition for conventional, stoichiometric SI combustion
- Isolate effects of entangled fuel properties, such as RON and HoV, on SI combustion and knock



- **Project Plans**

- Use the well-established CFR engine (identifies fuel RON and MON) as a research platform to develop scientific understanding of isolated fuel effects on SI combustion
- Investigate HoV effects on RON as a function of operating parameters (intake air temperature, compression ratio, and spark timing) for a range of fuels



collaborations established, CFR engine and test cell prep nearing completion

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ANL- Kolodziej/Ickes (2/2)

- **Progress**

- Collaboration established with Marathon Petroleum Corporation
- CFR engine refurbished and installation underway
- Test cell data acquisition installed
- Test matrix and analytical method defined



- **Deliverables**

- Completion of initial test matrix by Sep 30, 2016
- Boundary conditions of CFR engine for modeling to toolkit team
- Development of a two-parameter knock behavior characteristic which takes HoV into account is a potential outcome of this work



ORNL (Szybist) \$300k: fuel effects on the limits of EGR dilution for SI engines

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ORNL– Szybist (1/4)

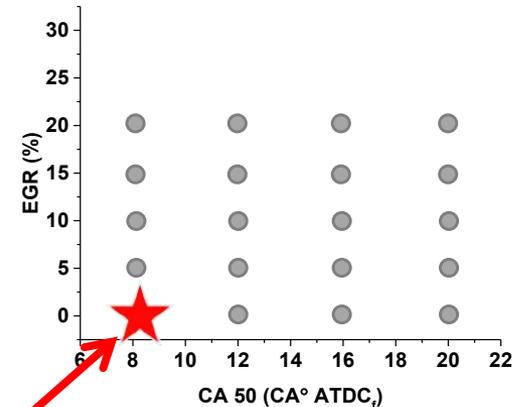
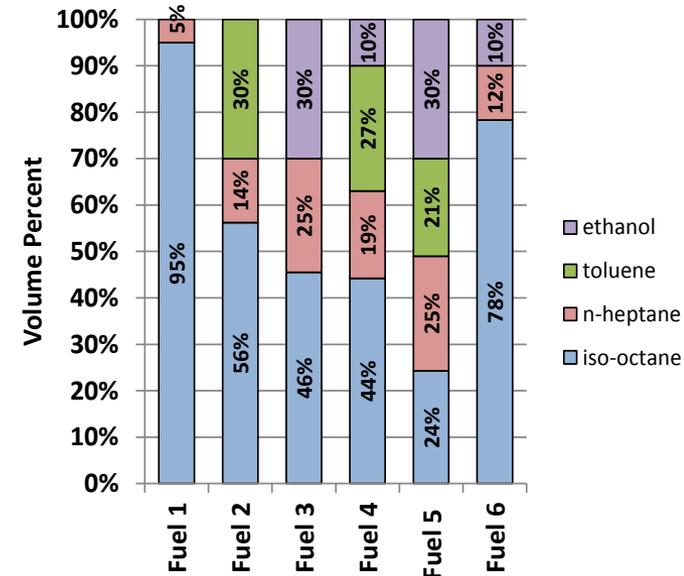
- **Objectives**

- Determine the magnitude and causes of fuel-specific differences in extending the EGR dilution limit

- **Methodology**

- Single cylinder version of GM Ecotec 2.0L, 9.2: CR
 - 2000 rpm, nominal load of 3.5 bar IMEP_g
- Side-mounted DI, laboratory air handling with external cooled EGR loop
- 6 fuel blends from pure components designed to vary flame speed, enable kinetic modeling
- Dilution limit defined by stability metric (COV) through EGR/combustion phasing space

- **Future Work:** Continuing work will be focused on high load dilution tolerance and pressure effects (>15 bar)

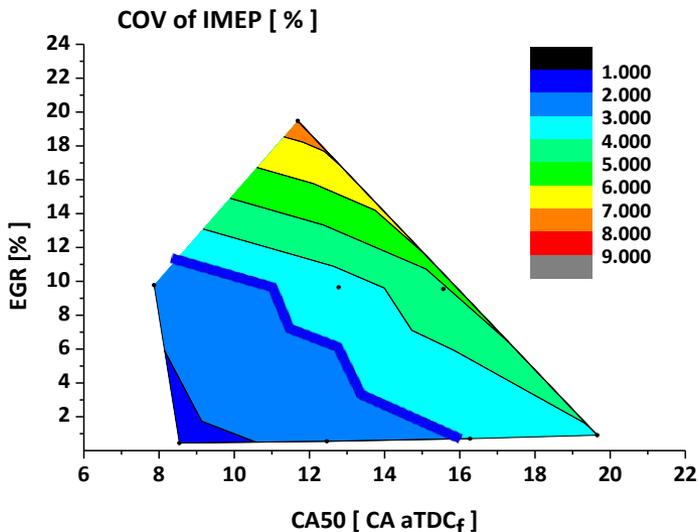


Baseline Operating Condition

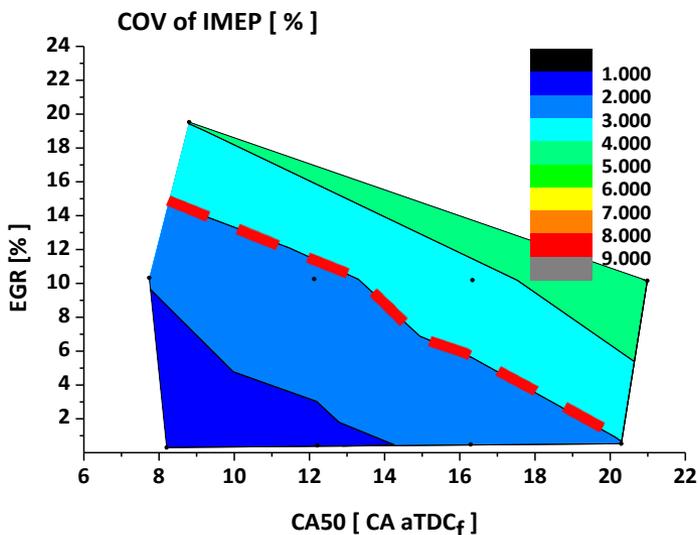


EGR dilution tolerance is defined by combustion stability, fuel specific differences are observed

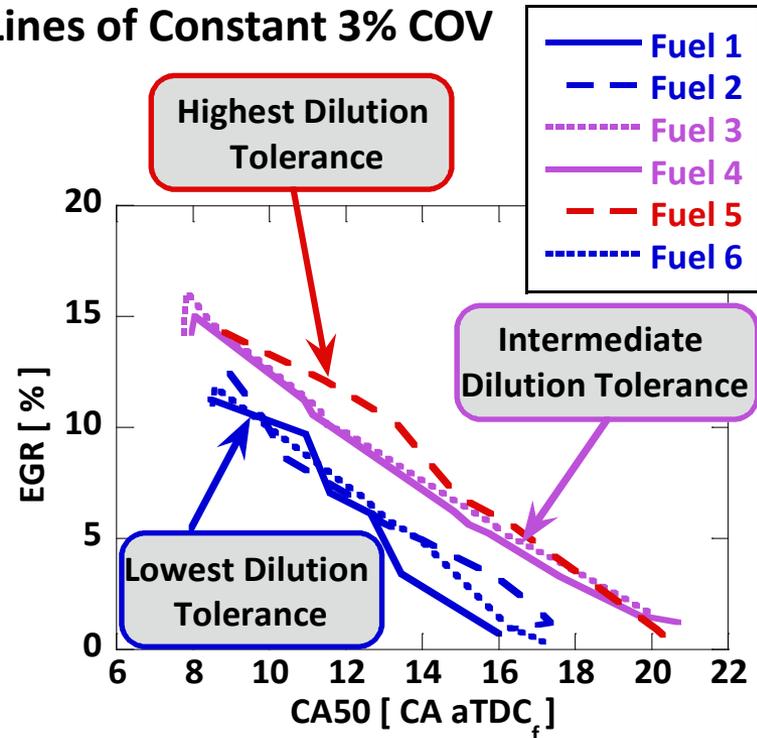
Fuel 1
95% iso-octane,
5% n-heptane



Fuel 5
24% iso-octane, 25% n-
heptane, 21% toluene,
30% ethanol



Lines of Constant 3% COV

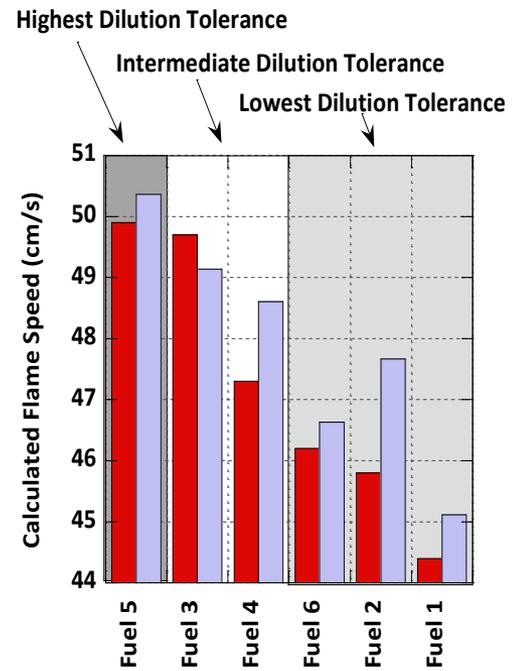


- Response to EGR is the similar for all fuels (similar slopes)
- Fuels-specific differences are due to different starting points

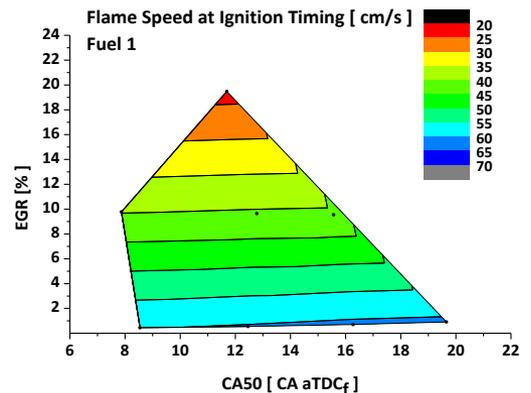


highest dilution tolerance fuels have highest flame speed

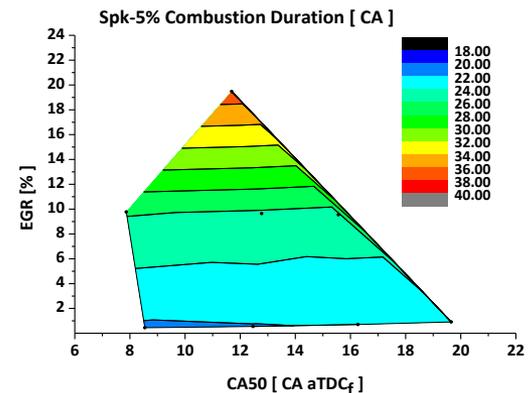
- Dilution tolerance correlates to laminar flame speed
 - Energy fraction mixing rule and kinetic calculations agree
 - Kinetics allow flame speed calculations at relevant temperature and pressure conditions
- Flame speed at ignition provides a good indication of spark-to-CA5, combustion stability
 - Experimental spark-to-CA5 to represents early flame growth
 - Rapid early flame kernel growth critical for combustion stability
 - Calculated flame speed in excellent agreement with experimental spark-to-CA5



Calculated Flame Speed at Ignition Timing is Insensitive to Spark Timing



Agrees with Experimental Results of Spark-to-CA5 duration



■ Mixing Rule (358 K)
■ Chemkin (348 K)

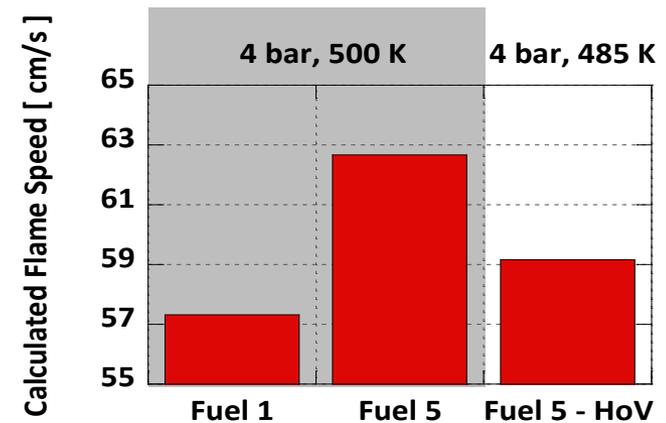
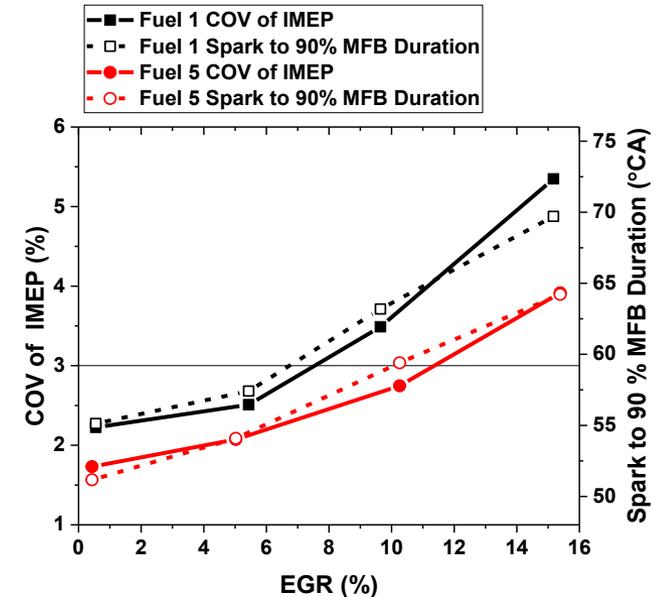


fuel-specific effects can be significant, HoV identified as possible dilution tolerance detractor

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ORNL-Szybist (4/4)

- Fastest flame speed fuel enables a 50% relative increase in EGR at a constant COV of 3% compared to the slowest flame speed fuel
 - 8% EGR to 12% EGR
 - Absolute EGR dilution tolerance will change with engine design and operating condition
 - Relative ranking between fuels is expected to remain for all homogeneous SI engines
- HoV may detract from dilution tolerance
 - Flame speed is a function of temperature
 - Up to 15 C of additional charge cooling expected for 30% ethanol
 - Flame speed calculations reveals this temperature reduction has a large impact on flame speed
- This project to move to high loads FY16 FY17
- ANL to investigate flame speed effects





ANL (Wallner) \$200k: fuel effects on EGR and lean dilution limits on SI combustion

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ANL- Kolodziej/Wallner (1/2)

Objectives

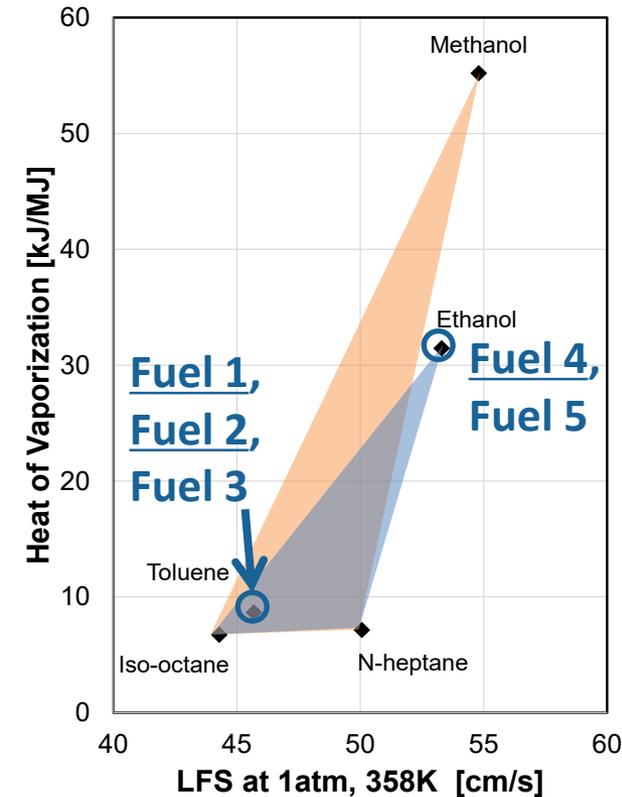
- Identify the influence of fuel properties on SI combustion lean and EGR dilution tolerance
- Quantify the relative impact of fuel properties on dilution tolerance compared to engine design parameters

Methodology

- Single cylinder version of Ford engine, 0.63L, 12.2:1 CR
 - 1500 rpm, loads of 3.2 and 5.6 bar IMEP, constant combustion phasing (8 CA aTDC_f)
- PFI and DI fueling

Project Plans

- Evaluate hypothesis that laminar flame speed (LFS), at nominal conditions, can be used to predict dilution tolerance (lean and EGR) of a fuel in SI combustion
- Determine the impact of HoV on dilution tolerance
- Compare fuel LFS lean/dilute tolerances against engine design parameters



Fuel Components	1	2	3	4	5
iso-octane		X	X	X	
n-heptane		X	X		X
toluene	X				
ethanol		X		X	
methanol			X		X



fuel matrix is isolating effects of LFS and HoV in preliminary data

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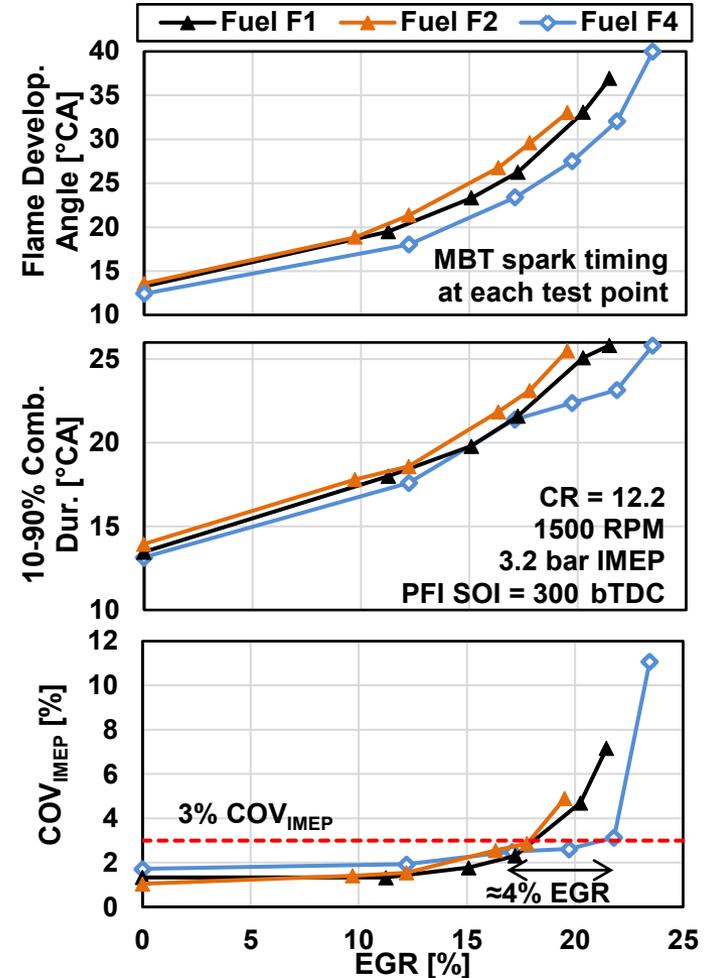
ANL- Kolodziej/Wallner (2/2)

• Progress

- Developed a method to identify fuel blends with target HoV and flame speed
- Derived fuel matrix and test program
- Preliminary results confirm positive correlation of flame speed with dilution tolerance

• Deliverables

- Comparison of fuel and engine design parameters on SI combustion lean dilution tolerance
- Completion of LFS hypothesis testing by June 30, 2016



Fuel F1: Low LFS, Low HOV (toluene)
Fuel F2: Low LFS, Low HOV (mixture)
Fuel F4: High LFS, High HOV (mixture)



ORNL (Splitter) \$150k: fuel effects in low speed pre-Ignition

LSPI promoting fuel, Fuel A SAE 2014-01-1226, RON =98.3
 LSPI suppressing fuel, Fuel K SAE 2014-01-1226, RON =99.4
 Haltermann EEE Tier II, RON =96.3
 Sunoco Optima, RON =98

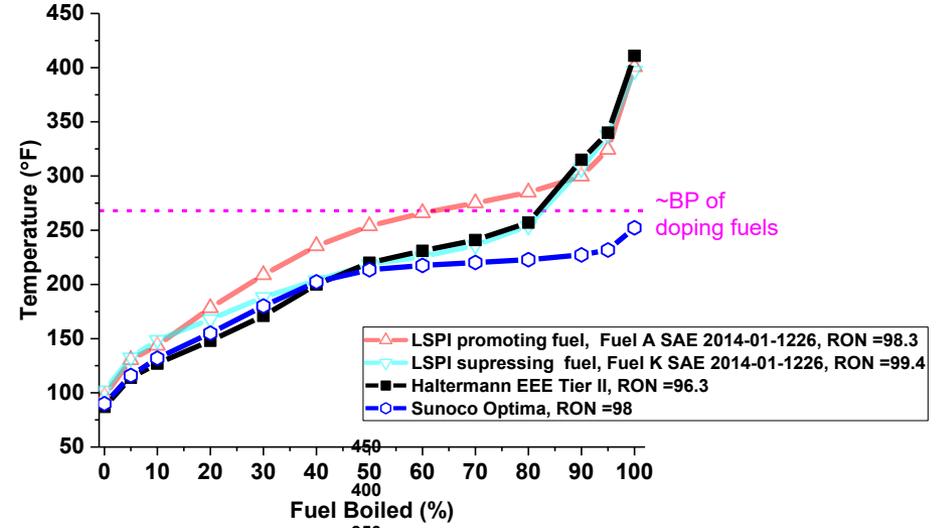
ORNL-Splitter (1/2)

Objectives

- To improve the understanding of fuel properties on low speed pre-ignition (LSPI)

Project Plans

- Single cylinder version of Ford 1.6L Ecoboost engine
 - Dedicated LSPI engine
 - Leverages lubricants LSPI activity
 - Automated LSPI test cycle
- Initial work focused on boiling point, HoV, and fuel structure on LSPI
 - Input to fuel merit function and co-optima down selection
- Extend focus to down selected co-optima fuels



Molecule Type	Bio-blendstock	Structure	BP (°C)	RON (-)	HoV (kJ/kg)	Planned
Alcohol	4-Methylpentan-3-ol	<chem>CC(C)C(O)CC</chem>	148	95.9	423	~BP of doping fuels
Alkane	(1S, 4S)-1,7,7-trimethylbicyclo[2.2.1]heptane	<chem>CC1(C)C2(C)CC1(C)C2</chem>	144.6	95.3	268	
Aromatic	o-Xylene	<chem>CC1=CC=C(C=C1)C</chem>	144	120	409	X
Ester	3-Methylbutyl acetate	<chem>CC(=O)OCC(C)C</chem>	142	100.6	291	X
Furan	2-methyl-5-propyl furan	<chem>CC1=CC=C(O1)CC</chem>	138	93.9	295	
Ketone	3-hydroxy-2-butanone	<chem>CC(=O)C(O)C</chem>	148	107.5	506	X



engine system is operational, on-track to meet Q4 project-level milestone

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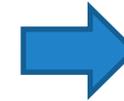
ORNL-Splitter (2/2)

FY 2016 Q2

• Progress

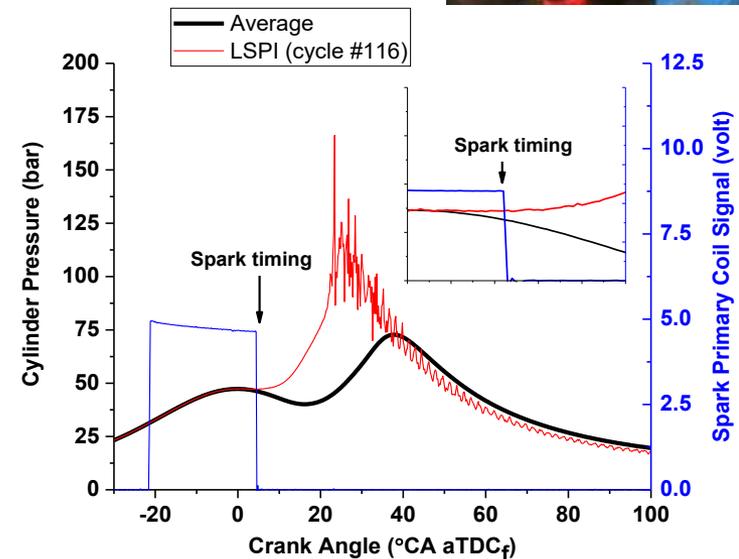
- Initial scoping fuels identified
- DAQ system collecting 15,000 consecutive cycles is operational through the leveraged lubricant project
- Engine failure occurred under LSPI conditions
 - New engine installed and operational

FY 2016 Q1



• Deliverables

- Initial molecular structure effects on LSPI tests Sept 30 2016
- Feed forward of findings into fuels merit function





numerous overarching and portfolio-level collaborations additional project-level collaborations

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- **Co-Optimization of Fuels and Engines** brings together expertise from across the National Laboratory system, working toward a common purpose. This effort has stakeholder engagement at a high level to ensure relevance.
 - 9 laboratories, engines, fuels, kinetics, simulation, biofuel development, LCA& TEA, market transformation
 - Monthly stakeholder engagement phone calls, industry listening days, external advisory board
- Projects presented at the semi-annual AEC program review meetings
- Engagement with ACEC Tech Team activities

Additional project-level collaborations with industry and academia

Sluder

Ford – Hardware and technical guidance
USDRIVE Fuels Working Group – multiple OEMs
and energy companies

Kolodziej/Ickes

Marathon Oil – Hardware, fuels, technical guidance

Szybist

FCA – Cody Baldwin-Squibb
University of Michigan – Yan Chang (student)

Kolodziej/Wallner

Ford - Hardware

Splitter

Driven Racing Oils – custom lubricants

Ratcliff

General Motors – Hardware and Technical Guidance



summary

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Relevance

Thrust I engine experiments are critical to understanding the role of fuel properties on efficiency. This information is critical to knowing how to value various fuel properties within “Co-Optima.”

Approach

Perform engine experiments that test the overarching “Co-Optima” fuel property hypothesis. Provide quantitative results that will aid in refining the Thrust I merit function. Interact with other teams within “Co-Optima” for modeling support, fuel selection, and fuels critical to the overall goal of reduced GHG emissions.

Accomplishments

- Once octane sensitivity has been accounted for, the role of HoV on KLSA is significantly diminished
- Multi-lab effort provided insight into the role of HoV as a thermal component to octane sensitivity
- EGR dilution tolerance is dominated by flame speed, related to early flame kernel growth

Collaborations

- “Co-Optima” has 9 National Labs, stakeholder engagement, and external advisory board
- Projects presented at AEC semi-annual program review, engaged with ACEC TT
- Numerous other project-level collaborations

Future Work

Continue to test the overarching “Co-Optima” fuel property hypothesis by incorporating biofuels with a wide range of chemical compositions into experiments, with support from modeling and kinetics efforts, in coordination with additional “Co-Optima” teams.